

Chapter 22

Hydrocarbon Compounds

22.1 Hydrocarbons

Organic Chemistry and Hydrocarbons

- Some thought that only living organisms could synthesize the organic compounds found in living cells.
- Synthetic chemical products (Friedrich Wöhler, 1828) show that organic molecules can be synthesized from inorganic compounds in nature.
- Today, organic chemistry involves the discovery of almost all carbon compounds, regardless of origin.

Introduction to Hydrocarbons

Hydrocarbon compounds containing only carbon and hydrogen.

Alkanes are the simplest hydrocarbon compounds.

Alkanes contain the most carbon chains in hydrocarbons, which always have the molecular formula:



Alkanes are the simplest and most inert molecules in organic compounds. However, their inert or stable character makes them the starting materials.

→ They undergo all reactions by first creating carbon-carbon bonds. In these reactions, they are the only carbon compounds.

The properties of hydrocarbons

Hydrocarbons are important molecules.

Hydrocarbons are the simplest compounds that are hydrocarbons, i.e., they hydrocarbons with low molecular weights and low boiling points.

Alkanes are the simplest hydrocarbons.

Alkanes are the simplest hydrocarbons, with the lowest boiling points.

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Puja Mehta

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