

Salary Sheet

COMPANY NAME

Sl. No.	Name of Employee	TOTAL GROSS SALARY								TOTAL SALARY	TOTAL DEDUCTIONS				
		Basic Salary	DA 50%	HRA 50%	Conveyance 5%	Medical Expenses 4%	Special 25%	Bonus 44%	TA 5%		Contribution to PF 12%	Provision on Tax	TDS	Salary Advance	TOTAL
1	Raj Sharma	30000	6000	15000	1500	1200	7500	13200	600	75000	18000				18000
2	Shradha Gaudh	15000	3000	7500	750	600	3750	6600	300	37500	18000			5000	46000
3	Danish D'Souza	10000	2000	5000	500	400	2500	4400	200	25000	14400				14400
4	Pooja Patel	12500	2500	6250	625	500	3125	5500	250	31250	18000				18000
5	Riya Patel	11000	2200	5500	550	440	2750	4840	220	27500	15840				15840
6	Joseph P.	7000	1400	3500	350	280	1750	3080	140	17500	10080				10080
7	Anand Patel	7500	1500	3750	375	300	1875	3300	150	18750	10800			5000	46800
8	Ganesh Raha	7000	1400	3500	350	280	1750	3080	140	17500	10080				10080
9	Vinod K. S.	7000	1400	3500	350	280	1750	3080	140	17500	10080			10000	110080
10	Divyans Kumar	4000	800	2000	200	160	1000	1760	80	10000	5760				5760
11	Shilpa R.	3740	748	1870	187	149.6	935	1643.6	74.8	9350	538.56				538.56
12	Sudha J.P.	4000	800	2000	200	160	1000	1760	80	10000	5760				5760
13	Deepika P.S.	4000	800	2000	200	160	1000	1760	80	10000	5760				5760
14	Lalit k r	6500	1300	3250	325	260	1625	2840	130	16250	9360				9360
15	Kavish K. M.	17500	3500	8750	875	700	4375	7700	350	43750	18000				18000

Sheet1

Sheet2

How To Prepare Salary Sheet In Excel

M Mosston



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study with quizlet and memorize flashcards containing terms like attractive force that acts between all nuclear particles that are extremely close together such as protons and neutrons in a nucleus stability of a nucleus depends on the band of stability and more

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jan 10 2017 nuclear transformation nuclear transformation reactions are nuclear reactions in which the nucleus of an element called target is bombarded with an accelerated particle called projectile bomb to transform the target into a new nucleus having a new chemical physical properties the projectiles bombs can be accelerated by using

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transformations can occur as a result of nuclear decay of radioactive isotopes or neutron capture neither procedure requires a change in the atom s element an atom gets transmuted into an atom of a different element when the number of protons in it is changed

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nuclear chemistry is the subfield of chemistry dealing with radioactivity nuclear processes such as nuclear transmutation and nuclear properties it is the chemistry of radioactive elements such as the actinides radium and radon together with the chemistry associated with equipment such as nuclear reactors which are designed to perform

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unlike a normal chemical reaction that forms molecules nuclear reactions result in the transformation of one element into another this property of nuclear reactions is used in nuclear power plants to gather nuclear energy the three common types of radiations are explained below

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a nuclear reactor consists of the following a nuclear fuel a fissionable isotope must be present in large enough quantities to sustain a controlled chain reaction the radioactive isotope is contained in tubes called fuel rods a moderator

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apr 3 2019 in order to understand the origin and character of individual radioactive emissions accompanying nuclear transformation processes we first need to answer three questions what is an unstable nucleus

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transmutation the changing of one chemical element into another transmutation involves a change in the nucleus or core of an atom and is therefore a nuclear reaction transformations may result from nuclear decay of radioactive isotopes or they can result from neutron capture

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in all accelerators the particles move in a vacuum to avoid collisions with gas molecules when neutrons are required for transmutation reactions they are usually obtained from radioactive decay reactions or from various nuclear reactions occurring in nuclear reactors

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nuclear reaction a nuclear reaction is a process in which two nuclei or a nucleus and an external subatomic particle collide to produce one or more new nuclides the nuclear reaction is a reaction that involves nucleoids the reactant nucleoid called the parent nucleoid usually transforms into a different nucleoid called the daughter

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nuclear transmutation is the conversion of one chemical element or an isotope into another chemical element 1 nuclear transmutation occurs in any process where the number of protons or neutrons in the nucleus of an atom is changed a transmutation can be achieved either by nuclear reactions in which an outside particle reacts with a nucleus

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jan 10 2021 write a balanced equation for each of the following nuclear reactions bismuth 212 decays into polonium 212 beryllium 8 and a positron are produced by the decay of an unstable nucleus neptunium 239 forms from the reaction of uranium 238 with a neutron and then spontaneously converts into plutonium 239

