Honors Chemistry 1st Semester Review Sheet – Ch 12

Concentration/Acid Chemistry

This packet, and all notes learned in class, will be helpful in preparing for the semester exam. Be sure to review all notes, homework, quizzes, and labs. You have also learned a lot about acids & bases this year. Use your textbook, your notes, and the internet to help you review. Be sure to begin studying well in advance of the exam. Do not wait until the night before to start studying!

Part I: Acids and Bases

Please be able to define or describe:

a) What does it mean to balance an equation?

b) How do we balance an equation? Be specific. Be sure to explain how to balance an equation with polyatomic ions.

c) Using the solubility rules you learned in class, determine if each of the following substances is soluble or insoluble in water:

Substance	Li_2CO_3	$Ca(OH)_2$	$Ba(NO_3)_2$	Al_2S_3	$Mg_3(PO_4)_2$	$NaCl$	$(NH_4)_2SO_4$
Soluble or Insoluble							

Write a balanced equation for each of the following reactions:

a) The decomposition of water

b) Combustion of propane

c) Decomposition of potassium chlorate

d) Synthesis of aluminum oxide

e) Decomposition of sodium hydrogen carbonate

f) Synthesis of magnesium nitride

g) Combustion of ethanol

h) Decomposition of ammonium carbonate

i) Synthesis of calcium phosphide

j) Combustion of butane

k) Decomposition of lithium hydroxide

l) Synthesis of sodium oxide

m) Combustion of methane

n) Decomposition of magnesium hydroxide

o) Synthesis of potassium chloride

p) Combustion of pentane

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Liliana Mammino,Jan Apotheker

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