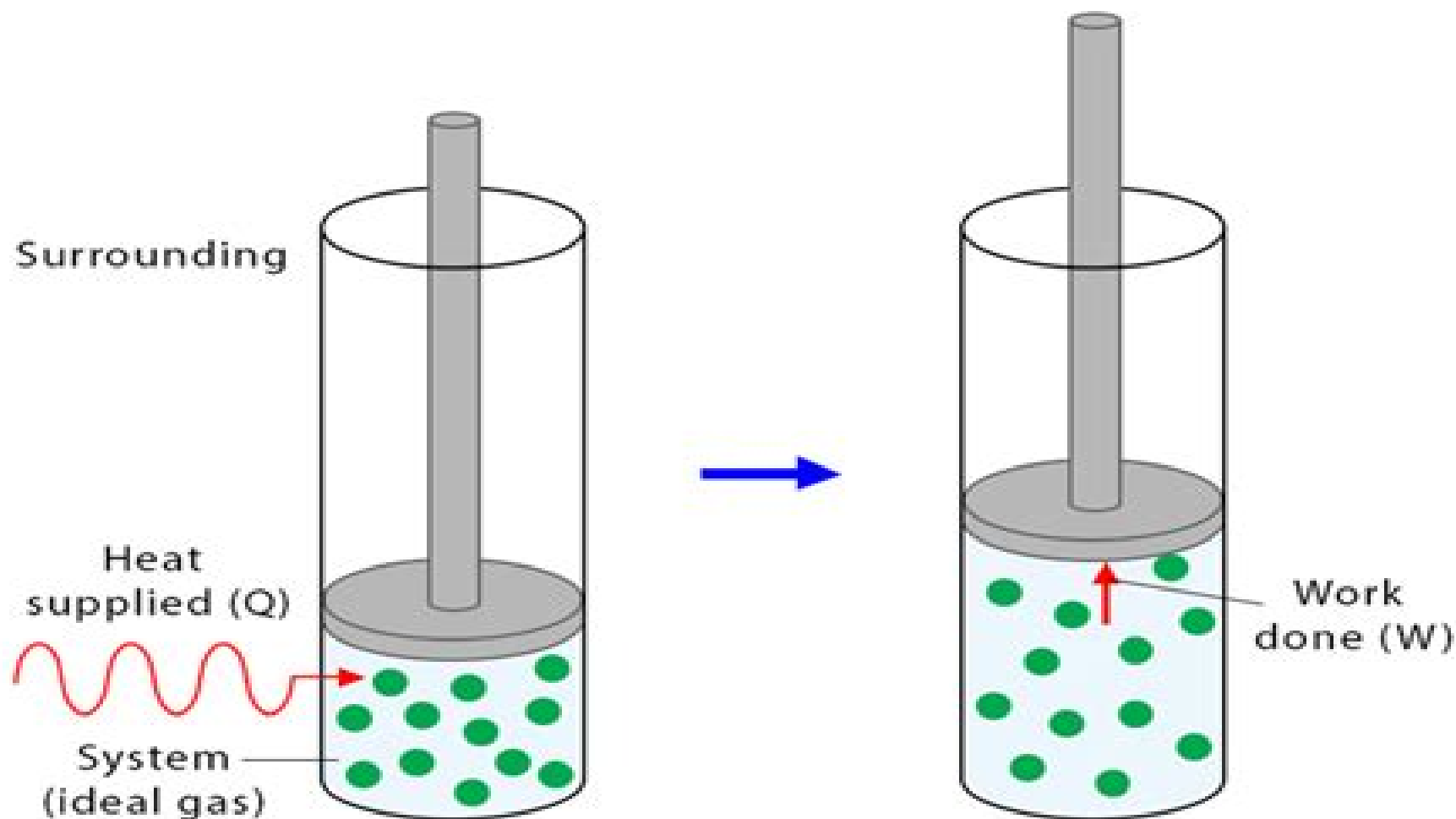


First Law of Thermodynamics

The change in internal energy (ΔU) of a system equals to the heat added to the system minus the work done



$$\Delta U = Q - W$$

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