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JS Bruner

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oh $\text{HCO}_2^- + \text{H}_4\text{O} \rightarrow \text{H}_2\text{O} + \text{H}_2\text{O}_2$ oxidation $4\text{Fe} + 8\text{H}^+ + \text{MnO}_4^- \rightarrow 4\text{Fe}^{2+} + 4\text{H}_2\text{O} + \text{Mn}^{2+}$ reduction $12\text{H}^+ + 5\text{CH}_3\text{OH} + 4\text{MnO}_4^- \rightarrow 5\text{HCO}_2^- + 4\text{H}_2\text{O} + 11\text{H}_2\text{O}$ d in acid

$\text{Cr}_2\text{O}_7^{2-}$