



GCSE Chemistry Titration Questions

GCSE 1-9 Science Chemistry Titration Question

An indicator is a substance that changes colour at the end point of a titration. It is used to determine when the reaction is complete. The colour change is called the end point.

Question - 1. Which of the following is NOT a suitable indicator for the titration of sodium hydroxide with ethanoic acid?

Options:

- A. Methyl orange
- B. Phenolphthalein
- C. Bromothymol blue
- D. Universal indicator

Answer: A

GCSE 1-9 Science Chemistry Titration Question

The student found that 25.0 cm³ of 0.100 mol dm⁻³ sodium hydroxide solution reacted with 25.0 cm³ of 0.100 mol dm⁻³ ethanoic acid. Calculate the concentration of sodium hydroxide solution in mol dm⁻³.

Options:

- A. 0.050
- B. 0.100
- C. 0.200
- D. 0.400

Answer: B

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The end point of a titration is reached when the indicator changes colour. This is the point at which the reaction is complete. The colour change is called the end point.

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Adam Robbins



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