

Potentiometry

Choose the correct answer:

No	Sentence	answer
1	All of the following are related to reference electrode except: a) of known and constant potential b) obeys Nernst equation. c) of high resistance to permit current of electric current. d) insensitive to activity of analyte solution.	C
2	Potential of standard hydrogen electrode is: a) Zero b) related to H^+ concentration c) 0.059 PH d) A and B e) All of the above	d
3	The increase in chloride ion concentration in Ag/AgCl electrode leads to: a) increase the potential b) decrease the potential c) no effect d) none of them	b
4	Potential of calomel electrode is proportional to chloride ion concentration. a) directly b) inversely c) equal d) no relation	b
5	In silver/silver chloride electrode, as KCl concentration increases: a) silver ions increases, potential increases. b) silver ions decreases, potential increases. c) silver ions decreases, potential remains the same. d) silver ions decreases, potential decreases.	d
6	In calomel electrode, as KCl concentration increases: a) mercury(I) ions increases, potential increases. b) mercury(I) ions decreases, potential increases. c) mercury(I) ions decreases, potential remains the same. d) mercury(I) ions decreases, potential decreases.	d
7	What is the electrode potential for a half-cell consisting of a cadmium electrode immersed in a solution that is 0.0150 M in Cd^{2+} ($E^0 = -0.403$ V) a) 0.457 b) 0.403 c) -0.457 d) -0.201	C

Mcq On Potentiometry

Chandresh Agrawal, Nandini Books



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Analytical Chemistry Modified Mr. Rohit Manglik,2024-03-01 EduGorilla Publication is a trusted name in the education sector committed to empowering learners with high quality study materials and resources Specializing in competitive exams and academic support EduGorilla provides comprehensive and well structured content tailored to meet the needs of students across various streams and levels

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