

1) Trend of main group elements when forming ions
Group 1 will lose 1 valence electron and become a 1+ CATION

*HIGHLIGHT NOTES FORM CW PACKET THAT SHOWS TRENDS FROM GROUP 1-8

- 2) Recognize polyatomic ion
- 3) Identify charges of main group elements
- 4) Label type of bond: Ionic bond or covalent bond or metallic bond
- 5) Draw Lewis structure of covalent compounds
- 6) Use electron-dot notation to demonstrate the formation of the ionic compound
- 7) Use VSEPR theory to predict and draw the molecular shape of molecules.

Main Group Elements Pogil

L Reisser



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