In the 18th century, several experiments led to the different LAWS we know of today.

- Joseph Louis Proust: Copper carbonate always had the same composition
- LAW OF DEFINITE PROPORTIONS or CONSTANT COMPOSITION: A compound always contains the same elements in certain definite proportions and in no other combinations.
- J.J.Berzelius: Prepared an extensive list of atomic weights; Lead sulfide experiments
- Henry Cavendish: 1783;
 - Hydrogen gas + Oxygen gas -> Water
- 1800: Volta designed a powerful battery W.Nicholson and A.Carlisle would use to separate water into its elements.

Law Of Definite Proportions Experiment Cupper Sulfide

Hosmer Ward Stone, Max Shaw Dunn

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