

The Hydrogen Atom – Student Challenge

Background Material

Classically, read the background information on Energy Levels, Light, and Transitions, and answer the following questions as clearly (and briefly) as possible.

Question 1: Compare the following table which compares how the Bohr Model and the Quantum Mechanical approach the Hydrogen atom. In what “interesting” facts describe things in the model you think are interesting that the other

Concept	Bohr Model	Quantum Model
Electrons	positively charged particles in flat orbits	positively charged particles in flat orbits
Electrons	orbiting electrons	negatively charged clouds
Orbitals	Electrons do not move, they give the probability picture	Classically, moving particles without orbits
Transitions of electrons to the first orbital	emitting a photon as the electron falls	Classical wave
Energy of the lowest level	$E_1 = -13.6 \text{ eV}$	It is about?

Question 2: Compare the following table which compares the properties of two different elements by adding their appropriate relationships. What does this accomplish for you? The next person has a harder time trying to understand whether frequency, wavelength, and the same relationship through space or within photon?

Hydrogen Atom Student Guide Solutions Naap

**Denise Guinn, Rebecca Brewer, Rachel
C. Lum**



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