

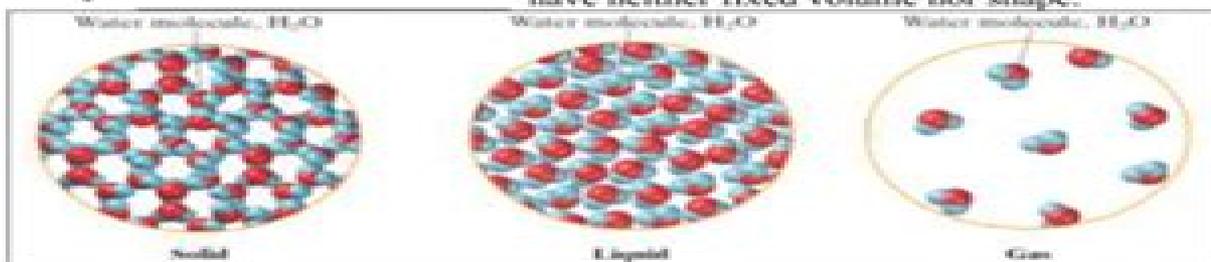
Holt Chemistry – Guided Notes, Chapter 1: The Science of Chemistry  
Section 1: What is Chemistry?

Objectives:

- **Describe** ways in which chemistry is a part of your daily life.
- **Describe** the characteristics of three common states of matter.
- **Describe** physical and chemical changes, and give examples of each.
- **Identify** the reactants and products in a chemical reaction.
- **List** four observations that suggest a chemical change has occurred.

A \_\_\_\_\_ is any substance that has a definite composition. A \_\_\_\_\_ is the process by which one or more substances change to produce one or more new substances. The \_\_\_\_\_ of \_\_\_\_\_ are the physical forms of matter which are solid, liquid, gas, and plasma.

- \_\_\_\_\_ have a fixed volume and shape that result from the way their particles are arranged.
- \_\_\_\_\_ have a fixed volume but not a fixed shape.
- \_\_\_\_\_ have neither fixed volume nor shape.



*Physical changes* are changes in which the \_\_\_\_\_ of a substance doesn't change.

- Changes of state are physical changes.

*Chemical changes* occur when the \_\_\_\_\_ of substances change and new substances form.

Evidence of a chemical change:

**a** When acetic acid, in vinegar, and sodium hydrogen carbonate, or baking soda, are mixed, the solution bubbles as carbon dioxide forms.

**b** When solutions of sodium sulfide and cadmium nitrate are mixed, cadmium sulfide, a solid precipitate, forms.

**c** When aluminum reacts with iron(III) oxide in the clay pot, energy is released as heat and light.

**d** When phenolphthalein is added to ammonia dissolved in water, a color change from colorless to pink occurs.

# Holt Chemistry Nuclear Chapter Test

**Brendan G. Carr**



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