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web how to balance hno 3 mg mg no 3 2 h 2 word equation nitric acid magnesium magnesium nitrate hydrogen gas type of chemical reaction for this reaction we have a single displacement reaction balancing strategies in this single displacement reaction the mg replaces the h in the hno3

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web nov 20 2020 in this video we ll balance the equation mg hno3 mg no3 2 no2 h2o and provide the correct coefficients for

each compound to balance $\text{Mg} + \text{HNO}_3 \rightarrow \text{Mg}(\text{NO}_3)_2 + \text{NO}_2 + \text{H}_2\text{O}$ you'll need to be

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web magnesium hydroxide nitric acid balanced total equation $\text{Mg}(\text{OH})_2(\text{aq}) + 2\text{HNO}_3(\text{aq}) \rightarrow \text{Mg}(\text{NO}_3)_2(\text{aq}) + 2\text{H}_2\text{O}$ total ionic equation

$\text{Mg}^{2+}(\text{aq}) + 2\text{OH}^{-}(\text{aq}) + 2\text{H}^{+}(\text{aq}) + 2\text{NO}_3^{-}(\text{aq}) \rightarrow \text{Mg}^{2+}(\text{aq}) + 2\text{NO}_3^{-}(\text{aq}) + 2\text{H}_2\text{O}(\text{l})$ net ionic equation $2\text{OH}^{-}(\text{aq}) + 2\text{H}^{+}(\text{aq}) \rightarrow 2\text{H}_2\text{O}(\text{l})$ sulfuric acid and lithium hydroxide

balanced total equation $\text{H}_2\text{SO}_4(\text{aq}) + 2\text{LiOH}(\text{aq}) \rightarrow \text{Li}_2\text{SO}_4(\text{aq}) + 2\text{H}_2\text{O}(\text{l})$