The Mystery Powders Lab: Physical and Chemical Changes

Oh no! This seems to be a new substance. It must be a chemical change.

Background Information: You have learned how to describe matter based on its physical and chemical properties. You have also learned some clues that tell you whether a physical change or a chemical change is occurring. Remember a physical change involves a change in a physical property but the substance does not change its identity. When a chemical change occurs a new substance is produced. There are hints like a color change or fizzing (a gas) that indicate a chemical change has taken place because a new substance is produced.

Objective: To determine the identity of four mystery powders which are similar in appearance by observing a physical property (solubility) and a chemical property (reactivity). You will also determine whether mixing these "mystery powders" with water, vinegar or iodine causes a physical or chemical change.

Materials:

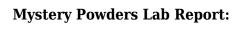
Procedure:

(Before starting get a good size piece of paper towel to use to clean the spatula in between stirring)

- Find the spatula labeled "A". Using the cut end of the spatula scoop up a very small amount of Powder "A" and place it in well #1. The powder should just cover the bottom.
- Place the same amount of Powder "A" in well #5 and well #9.
- Find the spatula labeled "B" for and place the same amount of Powder "B" in well # 2, 6, and 10.
- Find the spatula labeled "C" and place the same amount of Powder "C" in wells # 3,7,and 11.
- Find the spatula labeled "D". Place Powder "D" in wells # 4,8,and 12.

Mystery Powders Lab Report

JS Bruner



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