

WK	LSN	TOPIC/S-TOPIC	OBJECTIVES	L/ACTIVITIES	L/T AIDS	REFERENCE	REMARKS
1	1		<b>SCHOOL OPENING</b>				
	2	Structure of the atom and the periodic table - Structure of the atom	<b>By the end of the lesson, the learner should be able to:-</b> - Define atom - Name the subatomic particles and Where found - Describe the structure of the atom and label it	- Notes taking - Discussion - Drawing and labeling the atom	Chart of the structure of the atom	KLB BK 2 Pg 1	
	3	Atomic number and Atomic mass	<b>By the end of the lesson, the learner should be able to:-</b> - Define atomic number and atomic mass - Do simple calculations on how to find either of the given one of them	- Discussion - Notes taking - Calculation	Past questions	KLB BK 2 Pg 2-3	
	4	Isotopes	<b>By the end of the lesson, the learner should be able to:-</b> - Define isotopes - Name some elements that exhibit Isotopes - Name isotopes of hydrogen	- Discussion - Notes taking - Calculation	- Chalkboard - Charts	KLB BK 2 Pg 4	
2	1&2	Energy levels and Electron arrangement	<b>By the end of the lesson, the learner should be able to:-</b> - Write the electronic arrangement of The 1 <sup>st</sup> 20 elements - State the importance of the number of energy levels as well as the number of electrons in the outermost energy level	- Discussion - Notes taking - Writing of the E.C	Charts of E.C of the 1 <sup>st</sup> 20 elements	KLB BK 2 Pg 4-7	
	3  4	Development of the Periodic table	<b>By the end of the lesson, the learner should be able to:-</b> - Build up the periodic table for the 1 <sup>st</sup> 20 elements based on the no. of energy levels and no. of electrons in the outermost energy level. - Explain the position of an element in The table	- Discussion - Notes taking - Class activity	- Manila - Plasticine - Ruler - Felt pens - Charts of the 1 <sup>st</sup> 20 elements	KLB BK 2 Pg 8-9	
3	1&2	Relative atomic mass	<b>By the end of the lesson, the learner</b>				

# My Scheme Of Work For 3rd Term In Chemistry

**Puja Mehta**



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