

## JUNE 2010 – THE START OF THE NEW STYLE PAPER

### For 20 mark questions

Level	Mark	Descriptor
Level 1	1-7	Identification of points which are largely irrelevant.
Level 2	8-10	Typically, answers will consider 1 factor with 1 evaluative point or 2 points with no evaluation or list of several points with little analysis.
Level 3	11-14	Typically, answers will consider 2 factors with 1 evaluative point or 3 points with no evaluation.
Level 4	15-16	Typically, answers will consider 3 factors with 1 evaluative point.
Level 5	17-20	Typically, answers will consider 3 factors with at least 2 evaluative points.

### GUIDELINES FOR MARKING ESSAY QUESTIONS 6EC04 June 2012

Level	Part (a) Questions: Performance Criteria for Mark base 20	
Level 1	1-7	<ul style="list-style-type: none"> <li>Displays knowledge presented as facts without awareness of other viewpoints.</li> <li>Demonstrates limited understanding with little or no analysis.</li> <li>Attempts at selecting, applying, evaluating and presenting material which are irrelevant and unclear.</li> <li>None of the assessment objectives are covered satisfactorily.</li> </ul> <p>Identification of points which are largely irrelevant.</p>
Level 2	8-10	<ul style="list-style-type: none"> <li>Displays elementary knowledge of well learnt economic facts showing a generalised understanding together with limited analysis i.e. identification of points or a very limited discussion.</li> <li>Displays a limited ability to select and apply different economic ideas.</li> <li>Evidence presented has a basic relevance.</li> </ul> <p>Typically, answers will consider 1 factor with 1 evaluative point or 2 points with no evaluation or list of several points with little analysis.</p>
Level 3	11-14	<ul style="list-style-type: none"> <li>Displays knowledge and understanding of economic principles, concepts and theories as well as some analysis of issues i.e. answer might lack sufficient breadth and depth to be worthy of a higher mark.</li> <li>Shows some ability to apply economic ideas and relate them to economic problems.</li> <li>Employs different approaches to reach conclusions, presenting evidence with some relevance and coherence.</li> </ul> <p>Typically, answers will consider 2 factors with 1 evaluative point or 3 points with no evaluation.</p>
	11-12	Award in this range for 2 factors and 1 evaluative comment (or 3 factors with no evaluation) which are not well developed/reasoned.
	13-14	Award in this range for 2 factors and 1 evaluative comment (or 3 factors with no evaluation) which are well developed and reasoned.
Level 4	15-16	<ul style="list-style-type: none"> <li>Displays a good knowledge of economic principles, concepts and theories together with an analysis of the issues involved.</li> <li>Demonstrates an ability to select and apply economic ideas and to relate them to economic problems.</li> <li>Evidence of some evaluation of alternative approaches leading to conclusions which are presented in a relevant and coherent way.</li> </ul> <p>Typically, answers will consider 3 factors with 1 evaluative point.</p>
Level 5	17-20	<ul style="list-style-type: none"> <li>Displays a wide range of knowledge of economic principles, concepts and theories together with a rigorous analysis of issues.</li> <li>Demonstrates an outstanding ability to select and apply economic ideas to economic problems.</li> <li>Evaluation is well balanced and critical leading to valid conclusions.</li> <li>Material is presented in a relevant, clear and coherent way with evidence fully and reliably integrated.</li> <li>Excellence is displayed across all assessment objectives.</li> </ul> <p>Typically, answers will consider 3 factors with 2 evaluative points.</p>

# June 2013 B2 Markscheme Edexcel

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the molar concentration of the solute is sometimes abbreviated by putting square brackets around the chemical formula of

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web molarity is the number of moles of a substance per litre of solution also known as molar concentration a capital m signifies solutions labelled with molar concentration a 1.0 M solution contains 1 mole of solute per litre of solution molality is the number of solvent moles per kilogram

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web 1 molarity mv mass molar mass  $\times 0.4500 \text{ l} \times 0.825 \text{ g} \times 0.129 \text{ mol} \times 0.0129 \text{ m}$  2 molality  $0.825 \text{ g} \times 0.129 \text{ mol} \times 0.00581158 \text{ mol} \times 0.4500 \text{ kg} \times 0.129 \text{ m}$  3 mole fraction  $\frac{n_2}{n_1 + n_2} = \frac{0.825 \text{ g} \times 0.129 \text{ mol}}{0.825 \text{ g} \times 0.129 \text{ mol} + 0.00581158 \text{ mol} \times 450.0 \text{ g}}$  4 0.825 g 141.9579 g mol 0.00581158 mol h 2 o

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web apr 7 2014 molarity moles of solute liters of solution 8.422 a first convert 250 ml to liters  $250 \text{ ml} \times \frac{1 \text{ L}}{1000 \text{ ml}} = 0.25 \text{ L}$  then calculate molarity  $\frac{5 \text{ moles}}{0.25 \text{ L}} = 20 \text{ M}$  3 c a solution with molarity 2 requires 2 M of NaOH per liter so  $4 \times 2.8 \text{ M} = 11.2 \text{ M}$  4 a a solution of molarity 1.5 M requires 1.5 mol of NaOH to every litre of solvent

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web answer the correct option is c explanation the molarity of a mixture mmix can be calculated using the following formula  $M_{\text{mix}} = \frac{m_1}{V_1} + \frac{m_2}{V_2} + \frac{m_3}{V_3} + \dots$  1.5  $\times 480 \text{ L} + 2 \times 520 \text{ L} + 0.520 \text{ L} + 0.720 \text{ L} + 0.624 \text{ L} + 1.000 \text{ L} = 3.44 \text{ M}$  q5 with increase in temperature which of these changes a molality b fraction of solute present in water

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