

**CBSE Science - Grade X**  
**Solution for 2022-23 Examination**

Tooth enamel starts corroding in case the pH becomes acidic. The germs in the mouth decompose the food stuck in between the teeth and produce acids. These acids react with the enamel causing tooth decay. Tooth decay can be prevented by using toothpaste for brushing teeth and keeping oral hygiene.

**33.**

(a) Identify the reducing agent in the following reactions



(b) Define a redox reaction in terms of gain or loss of oxygen.

**Answer: (a)**

The substance that oxidized is the reducing agent and the substance which gets reduced is the oxidizing agent.



There is the oxidation of ammonia ( $\text{NH}_3$ ) and reduction of oxygen ( $\text{O}_2$ ) taking place so here  **$\text{NH}_3$  is the reducing agent** and  $\text{O}_2$  is oxidising agent.



There is the oxidation of  $\text{H}_2\text{O}$  and reduction of  $\text{F}_2$  taking place so here  **$\text{H}_2\text{O}$  is the reducing agent** and  $\text{F}_2$  is oxidising agent.



There is the oxidation of  $\text{CO}$  and reduction of  $\text{Fe}_2\text{O}_3$  taking place so here  **$\text{CO}$  is the reducing agent** and  $\text{Fe}_2\text{O}_3$  is oxidising agent.



There is the oxidation of  $\text{H}_2$  and reduction of  $\text{O}_2$  taking place so here  **$\text{H}_2$  is the reducing agent** and  $\text{O}_2$  is oxidising agent.

(b) Oxidation: A chemical reaction in which a substance gains oxygen is called oxidation.

Example:  $2\text{Cu} + \text{O}_2 = 2\text{CuO}$  ( $\text{Cu}$  is oxidized to  $\text{CuO}$ )

# March 2nengineering Science Question Paper

**RJ Shavelson**



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