

Lesson 1

WHAT IS FAITH?

What does the word "faith" mean? Write your own definition or copy a definition from a dictionary here:

FAITH:



The Bible gives its own definition of faith. Read Hebrews 11:1 and copy the verse here:

FAITH:

See if you can find and circle the word FAITH at least fifteen times in the puzzle below. The word is hidden twenty-six times! Look across, up and down, and diagonally.



Faith Activity Sheets

Clifford Lane



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web photoelectron spectroscopy electron configurations google classroom you might need periodic table using s p d f notation what is the electron configuration for a neutral atom of beryllium choose 1 answer $1s^2 2s^2$ a $1s^2 2s^2 2p^1$ b $1s^2 2s^2 2p^2$ c $1s^2 2p^2$ d $1s^2 2p^1$

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web elements in the second row of the periodic table place their electrons in the $2n$ shell as well as the $1n$ shell for instance lithium li has three electrons two fill the $1s$ orbital and the third is placed in the $2s$ orbital giving an electron configuration of $1s^2 2s^1$

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web one more point needs to be emphasized about the relationship between electron configuration and the periodic table the atoms of elements in the same vertical column of the table have similar electron configurations for example consider the alkaline earth elements group IIA

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web the electron removed during the ionization of beryllium $He\ 2s^2$ is an s electron whereas the electron removed during the ionization of boron $He\ 2s^2\ 2p^1$ is a p electron this results in a lower first ionization energy for boron even though its nuclear charge is greater by one proton

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web predictions from reliable sources have been used for these elements grayed out electron numbers indicate subshells filled to their maximum bracketed noble gas symbols on the left represent inner configurations that are the same in each period written out these are $He\ 2$ helium $1s^2$ $Ne\ 10$ neon $1s^2\ 2s^2\ 2p^6$

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web q1 what is the electron configuration of the element potassium answer the atomic number of potassium is 19 a neutral potassium element has 19 electrons its electron configuration is $1s^2\ 2s^2\ 2p^6\ 3s^2\ 3p^6\ 4s^1$ or simply put $Ar\ 4s^1$ q2 what is the total number of electrons that can be filled in s p and d subshells

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