

Chapter Test**Form B****Chapter 2**

Solve each equation. Then check your answer.

1. $x + 15 = 11$

2. $-4x + 7 = -21$

3. $k - 23 = -31$

4. $6w - 13 = 17$

5. $\frac{g}{6} = 9$

6. $9y = 63$

7. $7(x - 2) - 4x = 10$

8. $\frac{1}{3}(h - 7) = 9$

9. $\frac{3y - 3}{8} = \frac{3}{4}$

10. $9(m - 5) = 108$

11. $9a + 7 - 4a = -33$

12. $15.7y - 13.7y = -26$

13. If $-2d + 5 = -31$, what is the value of $3d - 9$?

14. Solve $6x - 7 = -19$. Justify each step.

Solve each formula in terms of the given variable.

15. $w = \frac{y + z}{3}$; z

16. $6x + 7y = 12$; y

Define a variable and write an equation to model each situation. Then solve.

17. The horticulture class is selling hanging baskets. There was originally \$432.10 in the class account. At the end of the sale, the class has \$1324.56 in its account. How much money did the class earn from the basket sale?
18. A local graphic designer charges \$25.00 for creating a T-shirt design and \$12.75 for each shirt printed with that design. If the math club has a bill totaling \$216.25 for T-shirts for its members, how many members purchased shirts?
19. Tickets to the movie theater for three adults and two children cost \$31.00. An adult's ticket costs \$4.50 more than a child's ticket. What is the cost of an adult's ticket?

Solve. If the equation is an identity, write *identity*. If it has no solution, write *no solution*.

20. $5(x + 2) = 5x - 2$

21. $3(d + 7) - d = 2(d + 10)$

22. $4(w - 3) - w = 3(w - 4)$

23. $8w - 10 + w = 14 + 10w + 4$

24. **Open-Ended** Write a word problem that could be solved using the equation $2(5 + w) = 42$.

Factoring Polynomials Chapter Test Form B Answers

Arturo Cuomo



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for the nitration 4 01 ml of nitric acid 4 00 ml of sulfuric acid and 4 56 ml bromobenzene were used to complete the reaction with these measurements bromobenzene was found to be the limiting reagent since only 4 56 ml of bromobenzene was used the amount of product predicted to be recovered was 8 7533 g

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in the nitration of bromobenzene a mixture of sulfuric acid and nitric acid react with one another nitric acid the weaker of the two acids will serve as base to form nitronium ion bromobenzene will attack the nitronium ion by nucleophilic attack

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then actual is your actual findings yield actual theoretical $\times 100$ what are the possible side reactions of nitration of bromobenzene dinitration can occur at ortho para or at both ortho positions occurs in overheating meta product happens very rarely because bromine is an ortho para director

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the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area theory and explanation of the nitration of bromobenzene if the temperature of the reaction mixture exceeds 60 °C

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the bromine substituent in bromobenzene is an ortho para director the para position is favored due to steric effects in the reaction that takes place in this lab a nitro group is substituted onto bromobenzene at the para position to form 1-bromo-4-nitrobenzene

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over 5 minutes slowly pipet this bromobenzene into the icy cold nitration mixture HNO_3 3 mL H_2SO_4 4 mL in the 125 mL flask and stir continue to stir the reaction mixture in the ice bath for 30 minutes do not heat in a new 250 mL erlenmeyer

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on the right the arenium cation has a nitro set with incredibly small 0.6 Å bond distances this will result in an

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the purpose of this lab was to observe where substituents are added on a benzene ring by way of electrophilic aromatic substitution the electrophilic source in the nitration of bromobenzene comes from a nitronium ion NO_2^+ which acts as the electrophilic species

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nitration of bromobenzene is an electrophilic aromatic substitution reaction the electrophile will be formed from 4 0ml of nitric acid and 4 0ml of sulfuric acid and will then react with 4 5ml of bromobenzene this will produce o bromonitrobenzene and p bromonitrobenzene

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