

A crystalline solid is formed when 2 or more atoms exchange valence electrons to increase stability.

Ionic bond is formed when valence electrons are transferred from one atom to another.

cations are atoms lose electrons to become [+] charged ions.

atoms gain electrons to become [-] charged.

In ionic compounds that ions are arranged in a crystal lattice. Strong forces hold the ions together.

Properties of ionic compounds:

- are high melting and boiling points
- are hard and easily crushed
- are conduct electricity when melted or dissolved because the ions are free to move

covalent bond is electrons are shared forming molecules

Covalent compounds have weak forces holding the molecules together.

Properties of covalent compounds:

- lower melting and boiling points
- Many covalent compounds are volatile liquids or gases.
- solids tend to crack
- are not conductors of electricity
- electronegativity is property that tells how strong an

Gbp Episode 6 answers

Peipei Pang



Gbp Episode 6 answers:

Whispering the Strategies of Language: An Mental Quest through **Gbp Episode 6answers**

In a digitally-driven world where screens reign great and instant interaction drowns out the subtleties of language, the profound strategies and emotional subtleties hidden within words frequently get unheard. However, located within the pages of **Gbp Episode 6answers** a charming fictional prize pulsing with natural thoughts, lies an exceptional journey waiting to be undertaken. Composed by a talented wordsmith, this charming opus attracts visitors on an introspective trip, softly unraveling the veiled truths and profound impact resonating within the material of each word. Within the psychological depths of this emotional review, we will embark upon a genuine exploration of the book is core themes, dissect their charming writing style, and fail to the strong resonance it evokes strong within the recesses of readers hearts.

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