

## Concept Review

### Section: How Substances Dissolve

1. Explain how you can speed up the dissolving process when preparing juice from frozen concentrate.  
\_\_\_\_\_  
\_\_\_\_\_
2. Explain why water is sometimes referred to as the universal solvent.  
\_\_\_\_\_  
\_\_\_\_\_
3. Draw a water molecule. Indicate the area where the electrons would be found. Label each atom and include all charges.  
\_\_\_\_\_  
\_\_\_\_\_
4. You make a salad dressing by mixing water, vinegar, olive oil, and seasonings. After a while, the oil separates and forms a layer above the other components.
  - a. Identify the polar and nonpolar liquids in the dressing.  
\_\_\_\_\_  
\_\_\_\_\_
  - b. Compare the strength of the attractions between a water molecule and a molecule of vinegar and between a water molecule and a molecule of olive oil.  
\_\_\_\_\_  
\_\_\_\_\_  
\_\_\_\_\_
5. Describe what happens when sodium chloride dissolves in water.  
\_\_\_\_\_  
\_\_\_\_\_  
\_\_\_\_\_

# Microsoft Office 2010 Concept Review Answers

**Siddappa N.Byrareddy**



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