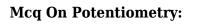
Potentiometry

Choose the correct answer:

No	Sentence	answer
1	All of the following are related to reference electrode except: a) of known and constant potential b) obeys Nernst equation. c) of high resistance to permit current of electric current. d) insensitive to activity of analyte solution.	с
2	Potential of standard hydrogen electrode is: a) Zero b) related to H* concentration c) 0.059 PH d) A and B e) All of the above	d
3	The increase in chloride ion concentration in Ag/AgCl electrode leads to: a) increase the potential b) decrease the potential c) no effect d) none of them	Ь
4	Potential of calomel electrode is proportional to chloride ion concentration. a) directly b) inversely c) equal d) no relation	ь
5	In silver/silver chloride electrode, as KCl concentration increases: a) silver ions increases, potential increases. b) silver ions decreases, potential increases. c) silver ions decreases, potential remains the same. d) silver ions decreases, potential decreases.	d
6	In calomel electrode, as KCI concentration increases: a) mercury(I) ions increases, potential increases. b) mercury(I) ions decreases, potential increases. c) mercury(I) ions decreases, potential remains the same. d) mercury(I) ions decreases, potential decreases.	d
7	What is the electrode potential for a half-cell consisting of a cadmium electrode immersed in a solution that is 0.0150 M in Cd2+ (E° = -0.403 V) a) 0.457 b) 0.403 C) -0.457 d) -0.201	c

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TD Snyder



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