Equilibrium Lab

Instructions: Complete all sections of this report and answer all questions in complete sentences for full credit.

Title: Equilibrium lab

Objective(s): Observe how changing the concentration of a component of a reaction effects its equilibrium

Hypothesis: Complete the following statements for your hypotheses. Choose one of the options in brackets by highlighting, bolding, or placing your answer in a different color.

If the amount of reactant increases, the rate of the forward reaction will [increase or decrease] and the rate of the reverse reaction will [increase or decrease].

If the amount of reactant decreases, the rate of the forward reaction will [increase or decrease] and the rate of the reverse reaction will [increase or decrease].

If the amount of product increases, the rate of the forward reaction will [increase or decrease] and the rate of the reverse reaction will [increase or decrease].

If the amount of product decreases, the rate of the forward reaction will [increase or decrease] and the rate of the reverse reaction will [increase or decrease].

Materials: For this lab, you are not expected to list the materials. But still notice the materials (lab equipment and chemicals) used in the lab to help you analyze the data, write the conclusion, and answer the post-lab questions.

Variables: List variables below as you work through the simulation.

Type	Variables
Controlled	
	equilibrium
Independent	how much of the reactants
Dependent	Transport Add
Substitution and the substitution of the subst	reaction rate

Procedure: Follow the steps described in the virtual lab. Record your observations and interpretations below.

Equilibrium reaction in the lab:

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Introducing Equilbrium Lab Answers

Frank M. Dunnivant, Elliot Anders

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Exploring General Chemistry in the Laboratory Colleen F. Craig, Kim N. Gunnerson, 2017-02-01 This laboratory manual is intended for a two semester general chemistry course The procedures are written with the goal of simplifying a complicated and often challenging subject for students by applying concepts to everyday life This lab manual covers topics such as composition of compounds reactivity stoichiometry limiting reactants gas laws calorimetry periodic trends molecular structure spectroscopy kinetics equilibria thermodynamics electrochemistry intermolecular forces solutions and coordination complexes By the end of this course you should have a solid understanding of the basic concepts of chemistry which will give you confidence as you embark on your career in science **An Inquiry-Based Introduction to Engineering Michelle** Blum, 2022-09-20 The text introduces engineering to first year undergraduate students using Inquiry Based Learning IBL It draws on several different inquiry based instruction types such as confirmation inquiry structured inquiry guided inquiry and open inquiry and all of their common elements Professor Blum's approach emphasizes the student's role in the learning process empowering them in the classroom to explore the material ask questions and share ideas instead of the instructor lecturing to passive learners about what they need to know Beginning with a preface to IBL the book is organized into three parts each consisting of four to ten chapters Each chapter has a dedicated topic where an initial few paragraphs of introductory or fundamental material are provided This is followed by a series of focused questions that quide the students learning about the concept's being taught Featuring multiple inquiry based strategies each most appropriate to the topic An Inquiry Based Approach to Introduction to Engineering stands as an easy to use textbook that guickly allows students to actively engage with the content during every class period Introduction to Intelligent Systems in Traffic and <u>Transportation</u> Ana L.C. Bazzan, Franziska Klügl, 2014-05-01 Urban mobility is not only one of the pillars of modern economic systems but also a key issue in the quest for equality of opportunity once it can improve access to other services Currently however there are a number of negative issues related to traffic especially in mega cities such as economical issues cost of opportunity caused by delays environmental externalities related to emissions of pollutants and social traffic accidents Solutions to these issues are more and more closely tied to information and communication technology Indeed a search in the technical literature using the keyword urban traffic to filter out articles on data network traffic retrieved the following number of articles as of December 3 2013 9 443 ACM Digital Library 26 054 Scopus and 1 730 000 Google Scholar Moreover articles listed in the ACM query relate to conferences as diverse as MobiCom CHI PADS and AAMAS This means that there is a big and diverse community of computer scientists and computer engineers who tackle research that is connected to the development of intelligent traffic and transportation systems It is also possible to see that this community is growing and that research projects are getting more and more interdisciplinary To foster the cooperation among the involved communities this book aims at giving a broad introduction into the basic but relevant concepts related to transportation systems targeting researchers and practitioners from computer science and information technology In addition the second part of the book

gives a panorama of some of the most exciting and newest technologies originating in computer science and computer engineering that are now being employed in projects related to car to car communication interconnected vehicles car navigation platooning crowd sensing and sensor networks among others This material will also be of interest to engineers and researchers from the traffic and transportation community **Research Based Undergraduate Science Teaching** Dennis W. Sunal, Cynthia S Sunal, Emmett L. Wright, Cheryl L. Mason, Dean Zollman, 2014-07-01 Research in Science Education RISE Volume 6 Research Based Undergraduate Science Teaching examines research theory and practice concerning issues of teaching science with undergraduates This RISE volume addresses higher education faculty and all who teach entry level science The focus is on helping undergraduates develop a basic science literacy leading to scientific expertise RISE Volume 6 focuses on research based reforms leading to best practices in teaching undergraduates in science and engineering The goal of this volume is to provide a research foundation for the professional development of faculty teaching undergraduate science Such science instruction should have short and longterm impacts on student outcomes The goal was carried out through a series of events over several years. The website at http nseus org documents materials from these events The international call for manuscripts for this volume requested the inclusion of major priorities and critical research areas methodological concerns and results of implementation of faculty professional development programs and reform in teaching in undergraduate science classrooms In developing research manuscripts to be reviewed for RISE Volume 6 researchers were asked to consider the status and effectiveness of current and experimental practices for reforming undergraduate science courses involving all undergraduates including groups of students who are not always well represented in STEM education To influence practice it is important to understand how researchbased practice is made and how it is implemented. The volume should be considered as a first step in thinking through what reform in undergraduate science teaching might look like and how we help faculty to implement such reform **Instructors Manual to Lab Manual** Ralph Petrucci, William Harwood, Geoffrey Herring, 2001 Introduction to Geomicrobiology Kurt O. Konhauser, 2009-03-12 Introduction to Geomicrobiology is a timely and comprehensive overview of how microbial life has affected Earth's environment through time It shows how the ubiquity of microorganisms their high chemical reactivity and their metabolic diversity make them a significant factor controlling the chemical composition of our planet The following topics are covered how microorganisms are classified the physical constraints governing their growth molecular approaches to studying microbial diversity and life in extreme environments bioenergetics microbial metabolic capabilities and major biogeochemical pathways chemical reactivity of the cell surface metal sorption and the microbial role in contaminant mobility and bioremediation biorecovery microbiological mineral formation and fossilization the function of microorganisms in mineral dissolution and oxidation and the industrial and environmental ramifications of these processes elemental cycling in biofilms formation of microbialites and sediment diagenesis the events that led to the emergence of life evolution of metabolic

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