

Characteristics of Gases

1. Gases or their mixtures are homogeneous in composition.
2. Gases possess high diffusibility.
3. Gases can be liquified, if subjected to low temperatures (below critical) or high pressures.
4. Gases have very low density due to negligible intermolecular forces.
5. Gases do not have definite shape and volume like liquids.
6. Gases undergo similar change with the change of temperature and pressure. In other words, gases obey certain laws known as gas laws.
7. Gases have infinite expansibility and high compressibility.
8. Gases have highest kinetic energy.
9. Gases exert pressure.
10. Gaseous molecules collide with one another and also with the walls of container with perfectly elastic collisions.
11. Thermal energy of gases $\gg$ molecular attraction.

Gaseous State Multiple Choice Questions

D Siedentop



Gaseous State Multiple Choice Questions:

Unveiling the Energy of Verbal Art: An Emotional Sojourn through **Gaseous State Multiple Choice Questions**

In a global inundated with monitors and the cacophony of fast communication, the profound power and emotional resonance of verbal art frequently diminish into obscurity, eclipsed by the continuous onslaught of sound and distractions. Yet, nestled within the musical pages of **Gaseous State Multiple Choice Questions**, a captivating work of fictional elegance that impulses with natural thoughts, lies an wonderful journey waiting to be embarked upon. Penned by a virtuoso wordsmith, this mesmerizing opus manuals visitors on an emotional odyssey, delicately exposing the latent possible and profound affect embedded within the elaborate web of language. Within the heart-wrenching expanse of this evocative examination, we will embark upon an introspective exploration of the book is main subjects, dissect their fascinating writing type, and immerse ourselves in the indelible impact it leaves upon the depths of readers souls.

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