

13. Complete the following table for the three experiments in Model 2.

|                        | Experiment C        | Experiment D                                     | Experiment E                  |
|------------------------|---------------------|--------------------------------------------------|-------------------------------|
| Independent Variable   | # of particles      | Temp.                                            | Internal<br>External Pressure |
| Dependent Variable     | Volume              | Volume                                           | Volume                        |
| Controlled Variable(s) | Temp / I+E pressure | External + Internal<br>Pressure / # of particles | Temp / # of particles         |

14. Provide a molecular level explanation for the increase in volume among the balloons in Experiment C. (How often and/or how hard are the molecules hitting the sides of the container?)

# of molecules  $\uparrow$ , so number of collisions  $\uparrow$  + Volume along with it

collisions  $\uparrow$  energy stays the same

15. Provide a molecular level explanation for the increase in volume among the balloons in Experiment D.

Temperature  $\uparrow$  increasing the kinetic energy of the particles as well  
which  $\uparrow$  collisions thus  $\uparrow$  pressure

16. Provide a molecular level explanation for the increase in volume among the balloons in Experiment E.

if external forces decrease then it allows internal forces  $\uparrow$   
thus  $\uparrow$  volume

# Gas Variables Pogil

**AW Chickering**



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