

E1

Volume = 1 unit

External pressure = 1 atm

Internal pressure = 1 atm

Temperature = 200 K

E2

Volume = 2 units

External pressure = 0.50 atm

Internal pressure = 0.50 atm

Temperature = 200 K

E3

Volume = 3 units

External pressure = 0.33 atm

Internal pressure = 0.33 atm

Temperature = 200 K

12. Consider the gas samples in Model 2.

- a. Name two materials that the containers in Model 2 could be made from that would ensure that they were "flexible"?

The container could be made of Rubber or Latex.

- b. What is always true for the external and internal pressures of a gas in a flexible container?

The external and internal pressures are equal in a flexible container

# Gas Variables Answers

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projects is reflected in modeling efforts      **Proceedings of the Ohio Gas Light Association** Ohio Gas Light Association,1904      **Trends in Oil and Gas Exploration** United States. Congress. Senate. Interior and Insular Affairs,1972

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*Gas World* ,1920      *Kaiparowits* ,1976

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