

12 • The Gas Laws**THE COMBINED GAS LAW**

In practical terms, it is often difficult to hold any of the variables constant. When there is a change in pressure, volume and temperature, the combined gas law is used.

$$\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2} \quad \text{or} \quad P_1 \times V_1 \times T_2 = P_2 \times V_2 \times T_1$$

$$K = ^\circ C + 273$$

*See
Scratch
paper*

Complete the following chart.

	P_1	V_1	T_1	P_2	V_2	T_2
1	1.50 atm	3.00 L	20.0°C 293 K	2.50 atm		30.0°C 303 K
2	720. torr	256. mL	25.0°C 298 K		250. mL	50.0°C 323 K
3	600. mmHg	2.50 L	22.0°C 325 K	760. mmHg	1.80 L	
4		750. mL	0.00°C 273 K	2.00 atm	500. mL	25.0°C 298 K
5	95.0 kPa	4.00 L		101. kPa	6.00 L	498.°C 471. K or
6	650. torr		100.°C 373 K	900. torr	225. mL	150.°C 423 K
7	850. mmHg	1.50 L	15.0°C 288 K		2.50 L	30.0°C 303 K
8	125. kPa	125. mL		100. kPa	100 mL	75.0°C 348 K

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Prentice Hall Chemistry, 2000 The Three Sisters Ying Wu, John J. Carroll, Yongle Hu, 2019-05-21 This is the seventh volume in the series Advances in Natural Gas Engineering focusing on carbon dioxide CO₂ capture and sequestration acid gas injection and enhanced oil recovery the three sisters of natural gas engineering This volume includes information for both upstream and downstream operations including chapters detailing the most cutting edge techniques in acid gas injection carbon capture chemical and thermodynamic models and much more Written by some of the most well known and respected chemical and process engineers working with natural gas today the chapters in this important volume represent the most state of the art processes and operations being used in the field Not available anywhere else this volume is a must have for any chemical engineer chemist or process engineer in the industry Advances in Natural Gas Engineering is an ongoing series of books meant to form the basis for the working library of any engineer working in natural gas today

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