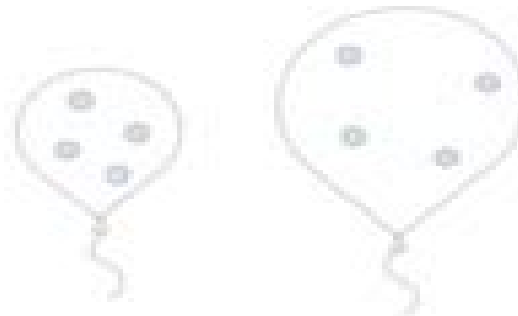


2) The relationship between temperature and volume:

a) Charles' law: At constant pressure, the volume of a fixed amount of gas is _____ proportional to its temperature in Kelvin; this means if the temperature increases, the volume _____.

b) Charles' law equation:

c) Particulate view:



d) Example 1: What would the final temperature of a gas be if 20 mL at a temperature of 40°C contracted to 5 mL?

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