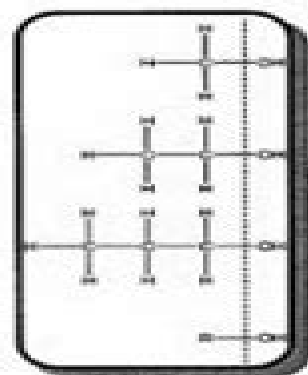


Naming Substituted Hydrocarbons

A substituted hydrocarbon is a hydrocarbon with an element other than hydrogen attached somewhere along the hydrocarbon chain. It is named in a similar fashion to a hydrocarbon. This can be illustrated with alcohols as an example. The compounds pictured to the lower left are



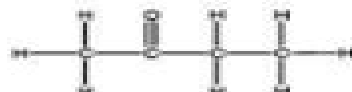
alcohols. They look like alkanes with $-OH$ at one end where a hydrogen would have been. The $-OH$ is called a functional group. The rest of the molecule is called a residue (R). The general formula for alcohols is $R-OH$. CH_3OH , the first alcohol pictured to the left is formed by substituting an $-OH$ group for hydrogen on methane (CH_4). As a result, it is called 1-methanol. The suffix *ol* shows that it is an alcohol. The root *methan* comes from methane. The number 1 shows the location of the $-OH$. The next alcohol in the series, CH_3CH_2OH , formed from ethane, is called 1-ethanol.

$CH_3CH_2CH_2OH$ is 1-propanol.

The alcohols and several other classes of substituted hydrocarbons are found in *Table R*. The root is determined by counting the number of carbons in the chain. For halides, the substitution is identified with a prefix. For the remaining substitutions, a suffix is used. (See *Table R*.) As with all hydrocarbons, the number and location of groups needs to be identified.



1,2,2-trifluoropropane



2-butanone

Name the following compounds using the rules for naming hydrocarbons and by referring to the reading and *Table R* above.

1. $CH_3CH_2CHOHCH_3$ _____

2. _____

Table R
Organic Functional Groups

Class of Compound	Functional Group	General Formula	Example
halide (halocarbon)	$-F$ (fluoro-) $-Cl$ (chloro-) $-Br$ (bromo-) $-I$ (iodo-)	$R-X$ (X represents any halogen)	$CH_3CHClCH_3$ 2-chloropropane
alcohol	$-OH$	$R-OH$	$CH_3CH_2CH_2OH$ 1-propanol
ether	$-O-$	$R-O-R'$	$CH_3OCH_2CH_3$ methyl ethyl ether
aldehyde	$\begin{array}{c} O \\ \\ -C-H \end{array}$	$R-\begin{array}{c} O \\ \\ C-H \end{array}$	$CH_3CH_2\begin{array}{c} O \\ \\ C-H \end{array}$ propanal
ketone	$\begin{array}{c} O \\ \\ -C- \end{array}$	$R-\begin{array}{c} O \\ \\ C-R' \end{array}$	$CH_3COCH_2CH_2CH_3$ 2-pentanone
organic acid	$\begin{array}{c} O \\ \\ -C-OH \end{array}$	$R-\begin{array}{c} O \\ \\ C-OH \end{array}$	$CH_3CH_2\begin{array}{c} O \\ \\ C-OH \end{array}$ propanoic acid
ester	$\begin{array}{c} O \\ \\ -C-O- \end{array}$	$R-\begin{array}{c} O \\ \\ C-O-R' \end{array}$	$CH_3CH_2COCH_3$ methyl propanoate
amine	$-N-$	$R'-N-R''$	$CH_3CH_2CH_2NH_2$ 1-propanamine
amide	$\begin{array}{c} O \\ \\ -C-NH \end{array}$	$R-\begin{array}{c} O \\ \\ C-NH \end{array}$	$CH_3CH_2\begin{array}{c} O \\ \\ C-NH_2 \end{array}$ propanamide

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web the ethiopian building code standard ebcs 2 1995 seems to have similar provisions for design of slender columns in sway frames with the american concrete institute aci however they have some clear differences one of these is the introduction of the substitute beam column frame in the ebcs 2 1995 for

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web fig 2 2 division of panels in flat slabs a column strip is a design strip with a width on each side of a column centerline equal to $0.25 l_x$ or if drops with dimension not less than $l_x/3$ are used a width equal to the drop dimension a middle strip is a

