

1. AIMS AND OBJECTIVES

- 1.1 The aim of this guide is to provide laboratories with guidance on best practice for the analytical operations they carry out. The guidance covers both qualitative and quantitative analysis carried out on a routine or non-routine basis. A separate guide covers research and development work (CITAC/EURACHEM Guide reference A1 on page 43).
- 1.2 The guidance is intended to help those implementing quality assurance in laboratories. For those working towards accreditation, certification, or other compliance with particular quality requirements, it will help explain what those requirements mean. The guidance will also be useful to those involved in the quality assessment of analytical laboratories against those quality requirements. Cross-references to ISO/IEC 17025, ISO 9000 and OECD Good Laboratory Practice (GLP) requirements are provided.
- 1.3 This document has been developed from the previous CITAC Guide 1 (which in turn was based on the EURACHEM/WEELAC Guide), and updated to take account of new material and developments, particularly the new requirements of the standard, ISO/IEC 17025.
- 1.4 This guide has been produced by a working group comprising David Holcombe, LGC, UK; Bernard King, NATA, Australia; Alan Squirrell, NATA, Australia and Maier Walsh, Suez Laboratory, Iceland. In addition, over the years leading to the drafting of this and earlier versions of the guide, there has been extensive input from a large number of individuals and organisations, including, CITAC, EURACHEM, EA, ILAC, AOAC, IUPAC, CCQM, and others (Refer Acronyms list on page 48).
- 1.5 This guide concentrates on the technical issues of quality assurance (QA), with emphasis on those areas where there is a particular interpretation required for chemical testing or related measurements. There are a number of additional aspects of QA where no guidance is given as these are fully addressed in other documents, such as ISO/IEC 17025. These include records; reports; quality systems; subcontracting; complaints; supplier's requirements; contract review; confidentiality and data handling.

2. INTRODUCTION

- 2.1 The value of chemical measurements depends upon the level of confidence that can be placed in the results. Increasingly, the chemical testing community is adopting QA principles which, whilst not actually guaranteeing the quality of the data produced, increases the likelihood of it being soundly based and fit for its intended purpose.
- 2.2 Appropriate QA can enable a laboratory to show that it has adequate facilities and equipment for carrying out chemical analysis and that the work was carried out by competent staff in a controlled manner, following a documented validated method. QA should focus on the key issues which determine quality results, costs and timeliness and avoid diversion of energies into less important issues.
- 2.3 Good QA practice, including its formal recognition by accreditation, certification etc., help to ensure that results are valid and fit for purpose. However, it is important for both laboratories and their customers to realise that QA cannot guarantee that 100% of the individual results will be reliable. There are two reasons for this:

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Table of Contents Eurachem Ea Guide 1

1. Understanding the eBook Eurachem Ea Guide 1
 - The Rise of Digital Reading Eurachem Ea Guide 1
 - Advantages of eBooks Over Traditional Books
2. Identifying Eurachem Ea Guide 1
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Eurachem Ea Guide 1
 - User-Friendly Interface
4. Exploring eBook Recommendations from Eurachem Ea Guide 1
 - Personalized Recommendations
 - Eurachem Ea Guide 1 User Reviews and Ratings
 - Eurachem Ea Guide 1 and Bestseller Lists

5. Accessing Eurachem Ea Guide 1 Free and Paid eBooks
 - Eurachem Ea Guide 1 Public Domain eBooks
 - Eurachem Ea Guide 1 eBook Subscription Services
 - Eurachem Ea Guide 1 Budget-Friendly Options
6. Navigating Eurachem Ea Guide 1 eBook Formats
 - ePub, PDF, MOBI, and More
 - Eurachem Ea Guide 1 Compatibility with Devices
 - Eurachem Ea Guide 1 Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Eurachem Ea Guide 1
 - Highlighting and Note-Taking Eurachem Ea Guide 1
 - Interactive Elements Eurachem Ea Guide 1
8. Staying Engaged with Eurachem Ea Guide 1
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Eurachem Ea Guide 1
9. Balancing eBooks and Physical Books Eurachem Ea Guide 1
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Eurachem Ea Guide 1
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Eurachem Ea Guide 1
 - Setting Reading Goals Eurachem Ea Guide 1
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Eurachem Ea Guide 1
 - Fact-Checking eBook Content of Eurachem Ea Guide 1
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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