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web gas stoichiometry practice problems using the following equation $2 \text{NH}_3(\text{g}) + \text{H}_2\text{SO}_4(\text{aq}) \rightarrow (\text{NH}_4)_2\text{SO}_4(\text{aq})$ calculate the volume of NH_3 in liters needed at 20°C and 25 atm to react with 150 kg of H_2SO_4

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web balancing equations and simple stoichiometry key balance the following equations $1 \text{N}_2 + 3 \text{F}_2 \rightarrow 2 \text{NF}_3$ $2 \text{C}_6\text{H}_{10} + 17 \text{O}_2 \rightarrow 12 \text{CO}_2 + 10 \text{H}_2\text{O}$ using the following equation to answer questions $8 \text{H}_2 + \text{C}_6\text{H}_8 \rightarrow 12 \text{CO}_2 + 10 \text{H}_2\text{O}$ key solutions for the stoichiometry practice worksheet