

Interpreting Chemical Formulas

(refer to Investigation 8-A, pp. 183-187)

Part 1: Formulas for Molecular Compounds

1. Read Part 1 in the text (pg. 183) and complete the table.

Name of Compound	Formula of Molecule	Elements Present	How Many Atoms of Each?	Total # of Atoms in Compound
	H_2O			
carbon monoxide				
carbon dioxide				
	C_2H_6			
	$\text{C}_2\text{H}_5\text{OH}$			
sulphuric acid	H_2SO_4			

Part 2: Formulas for Ionic Molecular Compounds

2. Read Part 2 in the text (pg. 185) and complete the following questions.

- How many sodium atoms are present in the model? _____
- How many chloride atoms are present in the model? _____
- The model has 1 Na atom for every _____ Cl atoms.
- The model has 1 Cl atom for every _____ Na atoms.
- The ratio of Na to Cl is _____ (pg. 185)
- NaCl is a common type of "salt". Write the formula for these other salts:
 - * potassium chloride _____
 - * lithium chloride _____

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Jiyuan Zhang



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