

Student Name: (Last, First, Middle Initial) _____, Section: _____, Professor's Name: _____, Date: _____

1. Dry ice (solid carbon dioxide) is also called "snakes." 1.000 grams of dry ice is placed in a 0.050 L container (that is evacuated) at 300.1 K. What is the pressure in the container after all of the dry ice has sublimated?

$$PV = nRT$$

$$\frac{1.000 \text{ g CO}_2}{44.01 \text{ g/mol}} \times \frac{1 \text{ mole CO}_2}{1.000 \text{ g}} = 0.02272 \text{ moles CO}_2$$

$$P = ?$$

$$V = 0.050 \text{ L (1 significant figure)}$$

$$n = 0.02272 \text{ moles CO}_2 \text{ (significant figures)}$$

$$R = 0.08206 \text{ L}\cdot\text{atm}/\text{mol}\cdot\text{K} \text{ (reference significant figures)}$$

$$T = 300.1 \text{ K} \rightarrow 300.1 \text{ K} \text{ (4 significant figures)}$$

$$P = \frac{nRT}{V} = \frac{(0.02272 \text{ moles}) (0.08206 \text{ L}\cdot\text{atm}/\text{mol}\cdot\text{K}) (300.1 \text{ K})}{0.050 \text{ L}} = 1.1 \text{ atm}$$

22 grams of a substance B, from the same source

- a) 1.000 atm
- b) 1.000 atm\cdothr
- c) 0.050 atm

2. A sample of chlorine gas is collected over a 0.050 L bottle at atmospheric temperature of pressure. How many moles of chlorine gas are in the container? How many grams?

$$0.050 \text{ L sample} \rightarrow 0.050 \text{ L}$$

$$\text{Molar Mass of Chlorine (molecular, or as a substance)} = 70.9 \text{ g/mol, grams}$$

How many moles

$$\frac{0.050 \text{ L}}{1} \times \frac{1 \text{ mole CO}_2}{22.4 \text{ L}} = 0.0022 \text{ moles CO}_2$$

$$\frac{0.050 \text{ L}}{1} \times \frac{1 \text{ mole CO}_2}{22.4 \text{ L}} \times \frac{70.9 \text{ g CO}_2}{1 \text{ mole CO}_2} = 0.77 \text{ g CO}_2$$

$$0.050 \text{ L (one (one) value) gas law, } PV = nRT$$

$$P = ? \text{ atm (reference significant figures)}$$

$$V = 0.050 \text{ L (1 significant figure)}$$

$$R = 0.08206 \text{ L}\cdot\text{atm}/\text{mol}\cdot\text{K} \text{ (reference significant figures)}$$

$$T = 273 \text{ K (reference significant figures)}$$

Gas Laws Extra Practice

Charles Henrickson



Gas Laws Extra Practice:

CliffsNotes Chemistry Practice Pack Charles Henrickson, 2010-02-08 About the Contents Pretest Helps you pinpoint where you need the most help Topic Area Reviews Measurement and Units of Measurement Matter Elements Compounds and Mixtures Atoms I The Basics Formulas and Names of Ionic Compounds Acids and Bases The Mole Elements and Compounds Percent Composition and Empirical and Molecular Formulas Chemical Reactions and Chemical Equations Calculations Using Balanced Equations Atoms II Atomic Structure and Periodic Properties Chemical Bonding The Formation of Compounds Gases and the Gas Laws The Forces between Molecules Solids and Liquids Solutions and Solution Composition Acids Bases and Neutralization Glossary Customized Full Length Exam Covers all subject areas Pretest that pinpoints what you need to study most Clear concise reviews of every topic Targeted example problems in every chapter with solutions and explanations Customized full length exam that adapts to your skill level **Physics I: 501 Practice Problems For Dummies (+ Free Online Practice)** The Experts at Dummies, 2022-05-10 Overcome your study inertia and polish your knowledge of physics Physics I 501 Practice Problems For Dummies gives you 501 opportunities to practice solving problems from all the major topics covered you Physics I class in the book and online Get extra help with tricky subjects solidify what you've already learned and get in depth walk throughs for every problem with this useful book These practice problems and detailed answer explanations will help you succeed in this tough but required class no matter what your skill level Thanks to Dummies you have a resource to help you put key concepts into practice Work through practice problems on all Physics I topics covered in school classes Step through detailed solutions to build your understanding Access practice questions online to study anywhere any time Improve your grade and up your study game with practice practice practice The material presented in Physics I 501 Practice Problems For Dummies is an excellent resource for students as well as parents and tutors looking to help supplement Physics I instruction Physics I 501 Practice Problems For Dummies 9781119883715 was previously published as Physics I Practice Problems For Dummies 9781118853153 While this version features a new Dummies cover and design the content is the same as the prior release and should not be considered a new or updated product *Certificate Physics Form 2* , **The Latest and Best of TESS** , 1991 **MCAT General Chemistry Review 2025-2026** Kaplan Test Prep, 2024-07-02 Kaplan's MCAT General Chemistry Review 2024 2025 offers an expert study plan detailed subject review and hundreds of online and in book practice questions all authored by the experts behind Kaplan's score raising MCAT prep course Prepping for the MCAT is a true challenge Kaplan can be your partner along the way offering guidance on where to focus your efforts and how to organize your review This book has been updated to match the AAMC's guidelines precisely no more worrying about whether your MCAT review is comprehensive The Most Practice More than 350 questions in the book and access to even more online more practice than any other MCAT general chemistry book on the market The Best Practice Comprehensive general chemistry subject review is written by top rated award winning

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Table of Contents Gas Laws Extra Practice

1. Understanding the eBook Gas Laws Extra Practice
 - The Rise of Digital Reading Gas Laws Extra Practice
 - Advantages of eBooks Over Traditional Books
2. Identifying Gas Laws Extra Practice
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Gas Laws Extra Practice
 - User-Friendly Interface
4. Exploring eBook Recommendations from Gas Laws Extra Practice
 - Personalized Recommendations
 - Gas Laws Extra Practice User Reviews and Ratings
 - Gas Laws Extra Practice and Bestseller Lists

5. Accessing Gas Laws Extra Practice Free and Paid eBooks
 - Gas Laws Extra Practice Public Domain eBooks
 - Gas Laws Extra Practice eBook Subscription Services
 - Gas Laws Extra Practice Budget-Friendly Options
6. Navigating Gas Laws Extra Practice eBook Formats
 - ePub, PDF, MOBI, and More
 - Gas Laws Extra Practice Compatibility with Devices
 - Gas Laws Extra Practice Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Gas Laws Extra Practice
 - Highlighting and Note-Taking Gas Laws Extra Practice
 - Interactive Elements Gas Laws Extra Practice
8. Staying Engaged with Gas Laws Extra Practice
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Gas Laws Extra Practice
9. Balancing eBooks and Physical Books Gas Laws Extra Practice
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Gas Laws Extra Practice
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Gas Laws Extra Practice
 - Setting Reading Goals Gas Laws Extra Practice
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Gas Laws Extra Practice
 - Fact-Checking eBook Content of Gas Laws Extra Practice
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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