

Gas Laws & SCUBA Diving

Read the accompanying article "Gas Laws and SCUBA Diving," ChemMatters, February 1993, pp.24.

Answer the following questions completely and upload to the assignment portal.

1. What is gas? name and describe?
2. What does diving "30 meters below" and "at different gas densities mean? Does being in a building 30 meters above sea level?
3. What gas is different? body and what affected by pressure changes?
4. What Boyle's Law
5. What does a SCUBA diver's lungs collapse as they ascend?
6. What would happen to a diver who ascends too quickly without having been a 30-meter diver? Explain in terms of Boyle's Law
7. What Henry's Law
8. What gas is associated with a warning/bubbles in the blood and at the body fluids?
9. What is another reason for decompression sickness?
10. Explain how these decompression in the blood and the symptoms of decompression sickness.
11. What is decompression?
12. What is gas law? explain why air is compressed (at least something off-GC) and what changes when that air is released to higher? Explain.
13. What relationships between the temperature of water and the solubility of a gas? is it an inverse relationship? Explain.
14. What gas law is the SCUBA diver's rule? is the decompression law is different from the other gas laws?

Gas Laws And Scuba Diving Chemmatters Answers

R Barnett



Gas Laws And Scuba Diving Chemmatters Answers:

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