

Gas Laws and Theory Quiz SCH3U

Name: _____

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(Product) 7 7 4 6

Assessment of Learning

$P = 8.31 \text{ kPa} \cdot \text{L} / \text{mol} \cdot \text{K} \times 1 \text{ mol} = 101.325 \text{ kPa} = 760 \text{ torr} = 760 \text{ mmHg}$ SATP = 25°C and 100 kPa

$\text{Mol} \times R$
100.3 kPa

STP = 0°C and

Multiple Choice. Circle the most correct answer. (780)

1. A pressure reading of 103.4 kPa is equivalent to _____ mmHg.
A) 653.4 B) 775.6 C) 774.6 D) 780.4

2. The particles that make up a solid move _____ than do the particles that make up a gas.

A) more quickly and farther C) more quickly
B) more slowly D) in the same way

3. According to the kinetic-molecular theory, how does a gas expand?

A) Its particles become larger. B) Collisions between particles become elastic.
C) Its temperature decreases. D) Its particles move greater distances apart.

4. If the temperature of a container of gas remains constant, how could the pressure of the gas increase?

A) The mass of the gas molecules increase B) The diffusion of the gas molecules increases.
C) The size of the container increases D) The number of gas molecules in the container increases.

5. Which of the following laws shows an inverse relationship between pressure and volume?

A) Boyle's Law B) Charles's Law C) Graham's Law D) Dalton's Law

6. A gaseous mixture at a total pressure of 1.50 atm contains equal molar amounts of He, Ne, and Ar. At constant temperature CO_2 gas is added to the mixture until the total pressure is 3.00 atm. Which of the following is a correct statement concerning partial pressures after the CO_2 addition?

A) The partial pressure of Ar has doubled.
B) The partial pressure of CO_2 is 3 times that of Ne

Gas Laws Quiz Answers

**Christopher R. Hren, John T.
Moore, Peter J. Mikulecky**



Gas Laws Quiz Answers:

Chemistry All-in-One For Dummies (+ Chapter Quizzes Online) Christopher R. Hren, John T. Moore, Peter J. Mikulecky, 2022-11-23 Everything you need to crush chemistry with confidence Chemistry All in One For Dummies arms you with all the no nonsense how to content you ll need to pass your chemistry class with flying colors You ll find tons of practical examples and practice problems and you ll get access to an online quiz for every chapter Reinforce the concepts you learn in the classroom and beef up your understanding of all the chemistry topics covered in the standard curriculum Prepping for the AP Chemistry exam Dummies has your back with plenty of review before test day With clear definitions concise explanations and plenty of helpful information on everything from matter and molecules to moles and measurements Chemistry All in One For Dummies is a one stop resource for chem students of all valences Review all the topics covered in a full year high school chemistry course or one semester of college chemistry Understand atoms molecules and the periodic table of elements Master chemical equations solutions and states of matter Complete practice problems and end of chapter quizzes online Chemistry All In One For Dummies is perfect for students who need help with coursework or want to cram extra hard to ace that chem test *Class 11-12 Chemistry MCQ (Multiple Choice Questions)* Arshad Iqbal, 2019-05-17 The Class 11 12 Chemistry Multiple Choice Questions MCQ Quiz with Answers PDF College Chemistry MCQ PDF Download Quiz Questions Chapter 1 6 Practice Tests with Answer Key 11th 12th Grade Chemistry Questions Bank MCQs Notes includes revision guide for problem solving with hundreds of solved MCQs Class 11 12 Chemistry MCQ with Answers PDF book covers basic concepts analytical and practical assessment tests Class 11 12 Chemistry MCQ PDF book helps to practice test questions from exam prep notes The Class 11 12 Chemistry MCQs with Answers PDF eBook includes revision guide with verbal quantitative and analytical past papers solved MCQs Class 11 12 Chemistry Multiple Choice Questions and Answers MCQs PDF Free download chapter 1 a book covers solved quiz questions and answers on chapters atomic structure basic chemistry chemical bonding chemistry experimental techniques gases liquids and solids tests for college and university revision guide Class 11 12 Chemistry Quiz Questions and Answers PDF free download eBook s sample covers beginner s solved questions textbook s study notes to practice online tests The book Grade 11 12 Chemistry MCQs Chapter 1 6 PDF includes college question papers to review practice tests for exams Class 11 12 Chemistry Multiple Choice Questions MCQ with Answers PDF digital edition eBook a study guide with textbook chapters tests for NEET MCAT GRE GMAT SAT ACT competitive exam College Chemistry Mock Tests Chapter 1 6 eBook covers problem solving exam tests from chemistry textbook and practical eBook chapter wise as Chapter 1 Atomic Structure MCQ Chapter 2 Basic Chemistry MCQ Chapter 3 Chemical Bonding MCQ Chapter 4 Experimental Techniques MCQ Chapter 5 Gases MCQ Chapter 6 Liquids and Solids MCQ The Atomic Structure MCQ PDF e Book Chapter 1 practice test to solve MCQ questions on Atoms atomic spectrum atomic absorption spectrum atomic emission spectrum molecules azimuthal quantum number Bohr s model Bohr s atomic model

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Table of Contents Gas Laws Quiz Answers

1. Understanding the eBook Gas Laws Quiz Answers
 - The Rise of Digital Reading Gas Laws Quiz Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Gas Laws Quiz Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Gas Laws Quiz Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Gas Laws Quiz Answers
 - Personalized Recommendations
 - Gas Laws Quiz Answers User Reviews and Ratings
 - Gas Laws Quiz Answers and Bestseller Lists
5. Accessing Gas Laws Quiz Answers Free and Paid eBooks

- Gas Laws Quiz Answers Public Domain eBooks
- Gas Laws Quiz Answers eBook Subscription Services
- Gas Laws Quiz Answers Budget-Friendly Options
- 6. Navigating Gas Laws Quiz Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Gas Laws Quiz Answers Compatibility with Devices
 - Gas Laws Quiz Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Gas Laws Quiz Answers
 - Highlighting and Note-Taking Gas Laws Quiz Answers
 - Interactive Elements Gas Laws Quiz Answers
- 8. Staying Engaged with Gas Laws Quiz Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Gas Laws Quiz Answers
- 9. Balancing eBooks and Physical Books Gas Laws Quiz Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Gas Laws Quiz Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Gas Laws Quiz Answers
 - Setting Reading Goals Gas Laws Quiz Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Gas Laws Quiz Answers
 - Fact-Checking eBook Content of Gas Laws Quiz Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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