Gas Laws: Combined Gas Law

The following problems may be solved using the relationships described by either Charles' or Boyles' laws, together in the combined gas law. Show all the steps to the solution of the problem.

 $V_1P_1/T_1 = V_2P_2/T_2$

 Carbon dioxide occupies a 2.54 dm³ container at STP. What will be the volume when the pressure is 150 KPa and 26°C?

 $V_3 = [2.54 \text{ dm}^3][101.3 \text{ KPa}][26^{\circ}\text{C} + 273\text{K}] = 1.88 \text{ dm}^3$ [150 KPa][273 K]

Oxygen occupies a fixed container of 5.5L at STP. What will happen to the pressure if the temperature rises to 300K?

P₂ = [5.5 L] [101.3 KPa][300 K] = 111.3 KPa [5.5 L] [273 K]

3. Methane is compressed in a closed 15.8 dm³ container at 101.3 KPa. If the volume drops to 8.7 dm³ and the temperature begins at 25°C and then drops to 18°C, what will the pressure of the gas be?

P₂ = [101.3 KPa][15.8 dm³][300 K] = 179.6 KPa [8.7 dm³][273 K]

4. A helium balloon is fully inflated at 1.2 L. When the clerk is filling the balloon, she stops to make sure it is not going to explode and checks the pressure. The pressure is 5.7 KPa and the temperature at the store is 24°C. The balloon is 0.75L full. She stops when the balloon is fully inflated and the pressure is 7.2 KPa. When the customer takes the balloon outside, it explodes. What was the temperature outdoors?

 $T = [24^{\circ}C + 273K][1.2 L][7.2 KPa] = 600 K$ [0.75 L][5.7 KPa]

5. Oxygen gas is added to a rebreather at the same rate as it would be in the open air. The volume is 4.8 dm³ and the temperature is 25 °C. The pressure is1 atm. Once the tank is lowered under the water the pressure increases to 2 atm and the temperature drops to 12°C. Is the tank in danger of exploding if it's maximum volume is 5.0 dm³?

 $V_2 = [4.8 \text{ dm}^3][1 \text{ atm}][12^{\circ}C + 273 \text{K}] = 2.1 \text{ dm}^3$, not in danger of exploding $[2 \text{ atm}][25^{\circ}C + 273 \text{K}]$

Ozone is formed due to electricity passing through the air and splitting some O₂ atoms
which then join briefly to form O₃, ozone. Your olfactory sense can detect 1.00 mL
ozone in 10 L of air at STP. You are surprised that the noreaster is accompanied by

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and underwater operations The general requirements of national defense and the specific requirements of underwater reconnaissance demolition ordnance disposal construction ship maintenance search rescue and salvage operations repeatedly give impetus to training and development Navy diving is no longer limited to tactical combat operations wartime salvage and submarine sinkings Fleet diving has become increasingly important and diversified since World War II A major part of the diving mission is inspecting and repairing naval vessels to minimize downtime and the need for dry docking Other aspects of fleet diving include recovering practice and research torpedoes installing and repairing underwater electronic arrays underwater construction and locating and recovering downed aircraft Handbook for Professional Divers Ronald M. **Gas Laws** ,2010 Mixed Gas Diving Tom Mount, Bret Gilliam, 1992-08 The Physics of Scuba Titcombe, 1973 **Diving** Marlow Anderson, 2011-06-28 The Physics of Scuba Diving features questions at the end of each chapter for which answers can be found by visiting http nup comphysics of scuba diving answer aspx GAS LAWS NARAYAN CHANGDER, 2024-04-01 Note Anyone can request the PDF version of this practice set workbook by emailing me at cbsenet4u gmail com You can also get full PDF books in quiz format on our youtube channel https www youtube com smartquiziz I will send you a PDF version of this workbook This book has been designed for candidates preparing for various competitive examinations It contains many objective questions specifically designed for different exams Answer keys are provided at the end of each page It will undoubtedly serve as the best preparation material for aspirants This book is an engaging guiz eBook for all and offers something for everyone This book will satisfy the curiosity of most students while also challenging their trivia skills and introducing them to new information Use this invaluable book to test your subject matter expertise Multiple choice exams are a common assessment method that all prospective candidates must be familiar with in today s academic environment Although the majority of students are accustomed to this MCQ format many are not well versed in it To achieve success in MCQ tests guizzes and trivia challenges one requires test taking techniques and skills in addition to subject knowledge It also provides you with the skills and information you need to achieve a good score in challenging tests or competitive examinations Whether you have studied the subject on your own read for pleasure or completed coursework it will assess your knowledge and prepare you for competitive exams guizzes trivia and more Simple Demonstrations of the Gas Laws William Maurice Dehn, 1908 Lesson 45 ADvTECH., The Gas Laws George I. Sackheim, 1964 Gases and Gas Laws Hobsons Publishing, PLC,1983-01-01 **The Gas Laws** Laurence Whisler, 1964 Gases Renée G. Ford 1962 Basic Diving Physics and Applications B. R. Wienke, 1994 Expanding on the Gas Laws ,2009-01-01 Inquiries in Science Chemistry Series Expanding on the Gas Laws Teacher's Guide Practical Application of Gas Laws William V. Wojciechowski, American Association for Respiratory Care, 1992 The Gas Laws. (A Harper & Row Programmed Text.). George Israel SACKHEIM,1964 Practical Application of Gas Laws Linda S. Slater, 1978

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