

The Gas Laws – Ch. 10 (p.313-322)

1. The gas left in a used aerosol can is at a pressure of 1 atm at 27°C. If this can is thrown into a fire, what is the internal pressure of the gas when its temperature reaches 927°C?

GIVEN	GAS LAW	WORK
T1= 300K P1= 101 kPa T2= 1200K P2= x	BOYLE	101 / 300 = x / 1200 x = 404
	FORMULA	
	P1/T1=P2/T2	
ANSWER: 404 kPa OR 4 ATM		

2. A sample of carbon dioxide occupies a volume of 3.50 L at 125 kPa. What pressure would the gas exert if the volume were decreased to 2.00 L?

GIVEN	GAS LAW	WORK
V1 = 3.50 L P1 = 125 kPa V2 = 2.00 L P2 = x	CHARLES	(125) (3.50) = (2.00) (x) x = 218.75 kPa
	FORMULA	
	P1V1 = P2V2	
ANSWER: 218.75 kPa		

3. A sample of propane occupies 250.0 L at 125 kPa and 38°C. Find its volume at 100.0 kPa and 95°C.

GIVEN	GAS LAW	WORK
v1 = 250.0 L P1 = 125 kPa T1 = 311 K V2 = x P2 = 100.0 kPa T2 = 368 K	COMBINED	(125) (250.0) / 311 = (100.0) (x) / 368 x = 369.77
	FORMULA	
	P1v1/T1=P2v2/T2	
ANSWER: 369.77 L		

Gas Law Worksheet 2 Answers

Roger Frost



Gas Law Worksheet 2 Answers:

A Guide to Teaching in the Active Learning Classroom Paul Baepler, J. D. Walker, D. Christopher Brooks, Kem Saichaie, Christina I. Petersen, 2023-07-03 While Active Learning Classrooms or ALCs offer rich new environments for learning they present many new challenges to faculty because among other things they eliminate the room's central focal point and disrupt the conventional seating plan to which faculty and students have become accustomed The importance of learning how to use these classrooms well and to capitalize on their special features is paramount The potential they represent can be realized only when they facilitate improved learning outcomes and engage students in the learning process in a manner different from traditional classrooms and lecture halls This book provides an introduction to ALCs briefly covering their history and then synthesizing the research on these spaces to provide faculty with empirically based practical guidance on how to use these unfamiliar spaces effectively Among the questions this book addresses are How can instructors mitigate the apparent lack of a central focal point in the space What types of learning activities work well in the ALCs and take advantage of the affordances of the room How can teachers address familiar classroom management challenges in these unfamiliar spaces If assessment and rapid feedback are critical in active learning how do they work in a room filled with circular tables and no central focus point How do instructors balance group learning with the needs of the larger class How can students be held accountable when many will necessarily have their backs facing the instructor How can instructors evaluate the effectiveness of their teaching in these spaces This book is intended for faculty preparing to teach in or already working in this new classroom environment for administrators planning to create ALCs or experimenting with provisionally designed rooms and for faculty developers helping teachers transition to using these new spaces

Simplified ICSE Chemistry Viraf J. Dalal, **Knowing Thermodynamics** Nicole Marie Gillespie, 2004, **The IT in Secondary Science Book** Roger Frost, 1994

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