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## SIMPLE CLASSIFICATION OF SUBSTANCES

1. 1995 Q 15

Explain how you would separate a mixture of nitrogen and oxygen gases given that their boiling points are  $-196^{\circ}\text{C}$  and  $-183^{\circ}\text{C}$  respectively. (2 marks)

.....

.....

.....

2. 1995 Q 8 P2

(a) The table below gives information about the major constituents of crude oil. Study it and answer the questions that follow.

Constituent	Boiling point ( $^{\circ}\text{C}$ )
Gases	Below $-90$
Petrol	$40-175$
Kerosene	$175-250$
Diesel oil	$250-350$
Lubricating oil	$350-400$
Bitumen	Above $400$

(i) Which one of the constituents of crude oil has molecules with the highest number of carbon atoms? (2 marks)

.....

.....

(ii) Name the process you would use to separate a mixture of petrol and diesel and explain how the separation takes place. (2 marks)

.....

.....

(iii) Explain why the constituent of crude oil and write its formula. (1 mark)

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**Lin, Cheng-Yao, Sun, Li**



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