

Quiz & Worksheet - Overview of Molecular Geometry

1. Which one of the following molecules would have tetrahedral molecular geometry?

- ☐ AsF_5
- ☐ CF_4
- ☐ XeF_4
- ☐ NH_3

2. The molecular geometry of a ClF_3 molecule is best described as:

- ☐ Trigonal pyramidal
- ☐ Trigonal planar
- ☐ Tetrahedral
- ☐ T-shaped

3. Based on the information in the lesson, which of the following molecules would have linear molecular geometry?

- ☐ HCN
- ☐ H_2S
- ☐ BF_3
- ☐ H_2CO

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Annelies Wilder-Smith

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