

Name: _____ DawgTag: _____

1. In the molecule methanol (COH_4), which atom will be in the center of the skeleton of the Lewis structure?

Carbon

2. What is the name given to multiple Lewis structures that describe the same molecule?

Resonance Structure

3. If two or more Lewis structures are possible to represent a molecule, what criterion is used to determine which structure or structures is preferred?

- smaller formal charge preferable to large.

- Like formal charge on adj atoms undesired (unless 0)

- more negative formal charge goes to more electronegative atom.

4. What is the shape of a molecule with three atoms bound and one lone pair of electrons on the central atom?

Trigonal Pyramidal

5. What is the shape of a molecule with the 2 lone pairs and 3 atoms bound to the central atom?

T-shaped

6. What is the shape of a CO_2 molecule?

Linear



7. What is the shape of a NH_3 molecule?

Trigonal Pyramid



8. If a central atom is making 5 bonds with 5 other atoms, what is the orbital hybridization of that central atom?

 sp^3d

9. If a central atom is making 3 bonds with 3 other atoms, what is the orbital hybridization of that central atom?

 sp^2

10. In the molecule methane, CH_4 , what is the H-C-H bond angle?

 109.5° 

Molecular Geometry And Bonding Pre Lab Answers

John T. Salinas



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Chemistry Eugene LeMay, Jr.,Herbert Beall,Karen M. Robblee,Douglas C. Brower,Prentice-Hall Staff,2002-02

Experimental Chemistry Robert J. Artz,1982 *Computer Based Projects for a Chemistry Curriculum* Thomas J. Manning,Aurora P. Gramatges,2013-04-04 This e book is a collection of exercises designed for students studying chemistry courses at a high school or undergraduate level The e book contains 24 chapters each containing various activities employing applications such as MS excel spreadsheets and Spartan computational modeling Each project is explained in a simple easy to understand manner The content within this book is suitable as a guide for both teachers and students and each chapter is supplemented with practice guidelines and exercises *Computer Based Projects for a Chemistry Curriculum* therefore serves to bring computer based learning a much needed addition in line with modern educational trends to the chemistry classroom

Exploring General Chemistry in the Laboratory Colleen F. Craig,Kim N. Gunnerson,2017-02-01 This laboratory manual is intended for a two semester general chemistry course The procedures are written with the goal of simplifying a complicated and often challenging subject for students by applying concepts to everyday life This lab manual covers topics such as composition of compounds reactivity stoichiometry limiting reactants gas laws calorimetry periodic trends molecular structure spectroscopy kinetics equilibria thermodynamics electrochemistry intermolecular forces solutions and coordination complexes By the end of this course you should have a solid understanding of the basic concepts of chemistry which will give you confidence as you embark on your career in science Prentice Hall Chemistry ,2000 **Biology** Joseph S.

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