

Chapter 22

Hydrocarbon Compounds

22.1 Hydrocarbons

Organic Chemistry and Hydrocarbons

- Know through what only being composed exclusively of the carbon and hydrogen atoms in their state
- Natural sources (petroleum, coal, etc.) are hydrocarbon substances are synthesized, organic organic compounds formed in nature
- Today, organic chemistry involves the discovery of almost all carbon compounds, except carbonates, cyanides, carbides, and oxides

Introduction to Hydrocarbons

Hydrocarbon compounds containing only carbon and hydrogen

Alkanes is an important class of hydrocarbons

Alkanes carbon from their molecular structure is defined when always from saturated bonds

Alkanes saturated hydrocarbons containing only carbon and hydrogen atoms, forming only single bonds, and are named as follows:



Alkanes is the largest and most varied category of hydrocarbon compounds. Alkanes from their molecular structure is defined when always from saturated bonds.

They always are formed by four carbon atoms, carbon chains. In this group that carbon chain length is defined by the number of carbon atoms.

The properties of hydrocarbons

Hydrocarbons are important molecules

Hydrocarbons from their molecular structure is defined when always from saturated bonds. In this group that carbon chain length is defined by the number of carbon atoms.

Alkanes are the largest and most varied category of hydrocarbon compounds.

Alkanes are the largest and most varied category of hydrocarbon compounds.

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Wenbin Ji

A decorative graphic consisting of a light blue horizontal bar with a rounded right end, and a red circular shape partially visible behind it.

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