

# Geometry

## Section 1.1 – 1.4 Review

Choose the best answer.

Refer to the figure for Exercises 1 and 2.



- Which represents the name of the ray whose endpoint is  $K$  and that passes through  $R$ ?  

A $\overrightarrow{RK}$	C $\overrightarrow{KS}$
B $\overrightarrow{KT}$	D $\overrightarrow{RK}$
- In the diagram, how many different rays have endpoint  $R$ ?  

F 1	H 3
G 2	J 4

Refer to the figure for Exercises 3 and 4.



- Which line contains points  $S$  and  $U$ ?  

A line $m$	C line $p$
B line $n$	D $\overleftrightarrow{ST}$
- In the diagram, how many different segments can be named?  

F 0	H 2
G 1	J 3

Refer to the figure for Exercises 5 and 6.



- What is  $MP$ ?  

A 1	C 4
B 2	D 5
- What is  $LP$ ?  

F -7.5	H 2.5
G -2.5	J 7.5

- $B$  is the midpoint of  $\overline{AC}$ .  $AB = 8v$ , and  $AC = 2v + 42$ . What is  $BC$ ?  

F 24	H 56
G 48	J 168
- An angle whose measure is  $70^\circ$  is what type of angle?  

A acute	C obtuse
B right	D straight
- $\overline{GJ}$  bisects  $\angle FGH$ ,  $m\angle FGJ = (7x - 9)^\circ$ , and  $m\angle HGJ = (2x + 36)^\circ$ . What is  $m\angle FGH$ ?  

F $43^\circ$	H $86^\circ$
G $54^\circ$	J $108^\circ$
- An angle measuring  $22^\circ$  is bisected. What is the measure of the angles that are formed?  

A $11^\circ$	C $33^\circ$
B $22^\circ$	D $44^\circ$
- Which angle forms a linear pair with  $\angle MPS$ ?



- |                |                |
|----------------|----------------|
| F $\angle RPN$ | H $\angle MPJ$ |
| G $\angle RPM$ | J $\angle MPK$ |
- If  $m\angle Q = (8x - 40)^\circ$ , what is the measure of its supplement?  

A $(130 - 8x)^\circ$	C $90^\circ$
B $(220 - 8x)^\circ$	D $180^\circ$

# Holt Geometry 2 Practice

**Edson Homer Taylor**

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

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web study with quizlet and memorize flashcards containing terms like identify the type of reaction represented by each equation a  $\text{H}_2 + \text{Cl}_2 \rightarrow 2\text{HCl}$  b  $\text{CH}_4 + 2\text{O}_2 \rightarrow \text{CO}_2 + 2\text{H}_2\text{O}$  which chemical symbols will complete the equation for this single displacement reaction  $2\text{NaBr} + \text{Cl}_2 \rightarrow$  balance  $\text{KClO}_3 \rightarrow \text{KCl} + \text{O}_2$  and more