

## Quarter 3 Exam Review

### Multiple Choice

Always, the choice that best completes the statement or answers the question.

- \_\_\_\_\_ 1. In a chemical reaction
  - a. the mass of the reactants equals the mass of the products.
  - b. the mass of the product is greater than the mass of reactants.
  - c. the material atoms in the reactants and products must change.
  - d. energy is lost must be added to the reactants.
- \_\_\_\_\_ 2. When a ratio, provided by a chemical reaction, compares facts the solution it is called
  - a. a percent.
  - b. a constant.
  - c. a percent.
  - d. the mass of the product.
- \_\_\_\_\_ 3. In writing a chemical equation that produces hydrogen gas, the correct representation of hydrogen gas is
  - a. H
  - b. H<sub>2</sub>
  - c. H<sub>2</sub>
  - d. H<sub>2</sub>
- \_\_\_\_\_ 4. To balance a chemical equation, it may be necessary to adjust the
  - a. coefficients.
  - b. subscripts.
  - c. formulas of the products.
  - d. masses of products.
- \_\_\_\_\_ 5. According to the law of conservation of mass, the total mass of the reacting substances is
  - a. always less than the total mass of the products.
  - b. always less than the total mass of the products.
  - c. sometimes more and sometimes less than the total mass of the products.
  - d. always equal to the total mass of the products.
- \_\_\_\_\_ 6. In the chemical equation  $2\text{Mg}(s) + \text{O}_2(g) \rightarrow 2\text{MgO}(s)$ 
  - a. Mg represents the product magnesium.
  - b. Mg represents the reactant magnesium.
  - c. the atomic yield magnesium.
  - d. O<sub>2</sub> represents the product oxygen gas.
- \_\_\_\_\_ 7. When the reaction  $2\text{FeCl}_3 + 3\text{H}_2 \rightarrow 2\text{Fe} + 6\text{HCl}$  is correctly balanced, what is the coefficient of Fe?
  - a. 4
  - b. 4
  - c. 6
  - d. 6
- \_\_\_\_\_ 8. Which coefficients correctly balance the formula equation  $\text{CaCl}_2 + \text{H}_2\text{O} \rightarrow \text{Ca}(\text{OH})_2$ ?
  - a. 1, 1, 2
  - b. 1, 2, 1
  - c. 1, 2, 1
  - d. 1, 1, 1
- \_\_\_\_\_ 9. In which kind of reaction do two or more substances combine to form a new compound?
  - a. decomposition reaction
  - b. single reaction
  - c. double displacement reaction
  - d. synthesis reaction
- \_\_\_\_\_ 10. The equation  $\text{A} + \text{B} \rightarrow \text{C} + \text{D}$  is the general equation for a
  - a. synthesis reaction.
  - b. decomposition reaction.
  - c. combination reaction.
  - d. single displacement reaction.

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**L Reisser**



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