

## CHEM EXAM 3 Fall 2009 - Form A

Fill in your name, CID and section on this test booklet. Fill in and bubble in your name, CID (for the letter "C" bubble zero), section and "001" on the scantron form. For questions #66, mark A. For each question, read all possible choices then mark the best one on your scantron. Use a #2 pencil and make clean answers to insure proper scoring. When finished, turn in your scantron and show this booklet and your CID to the proctor. All questions are worth 4 points each. Let your knowledge show!

1. Which of the following is balanced correctly?

- A.  $\text{CH}_4 + 2 \text{O}_2 \rightarrow \text{CO}_2 + 2 \text{H}_2\text{O}$   
 B.  $6 \text{H}_2\text{O} + \text{P}_4\text{O}_{10} \rightarrow 4 \text{H}_3\text{PO}_4$   
 C.  $3 \text{H}_2 + 3 \text{N}_2 \rightarrow 2 \text{NH}_3$   
 D.  $2 \text{C}_2\text{H}_6\text{O} + 3 \text{O}_2 \rightarrow 2 \text{CO}_2 + 4 \text{H}_2\text{O}$

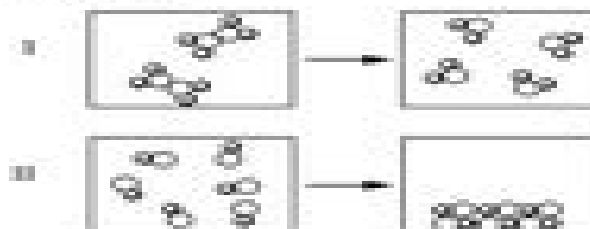
A chemical equation is balanced when the same number of atoms are in the reactants and products. The reactants in B contain 16 oxygen atoms, but the products contain only 12 (B). The reactants in C contain 6 nitrogen atoms, but the product contains only 2 (C). The reactants in D contain 6 carbon atoms, 16 hydrogen atoms, and 8 oxygen atoms. The products contain 2 carbon atoms, 8 oxygen atoms, and 8 hydrogen atoms (D).

2. Which of the following is an oxidation-reduction reaction?

- A.  $2 \text{NaCl (aq)} + \text{Ca(NO}_3)_2 \text{ (aq)} \rightarrow 2 \text{NaNO}_3 \text{ (aq)} + \text{CaCl}_2 \text{ (aq)}$   
 B.  $\text{CrCl}_3 \text{ (aq)} + 3 \text{KOH (aq)} \rightarrow 3 \text{KCl (aq)} + \text{Cr(OH)}_3 \text{ (s)}$   
 C.  $\text{KOH (aq)} + \text{HCl (aq)} \rightarrow \text{KCl (aq)} + \text{H}_2\text{O (l)}$   
 D.  $2 \text{CrCl}_3 \text{ (aq)} + \text{CuCl}_2 \text{ (aq)} \rightarrow 2 \text{CrCl}_2 \text{ (aq)} + \text{Cu (s)}$

An oxidation-reduction reaction is one in which electrons are transferred from one or more atoms in the reactants to one or more atoms in the products. The reactants and the products in A are all strong electrolytes (a net ionic equation, A). Two aqueous solutions react in B to form a solid precipitate, B. A strong acid reacts with a strong base in C to form an ionic salt and water (acid-base, C). Two aqueous solutions react to form a solid in D, but chromium in  $\text{CrCl}_3$  has an oxidation number of +3. In  $\text{CrCl}_2$ , it has an oxidation number of +2. Copper in  $\text{CuCl}_2$  has an oxidation number of +2. Solid copper has an oxidation number of 0. Therefore, chromium is oxidized while copper is reduced (an oxidation-reduction reaction).

3. Which of the following transfers heat to the surroundings?



- A. I only  
 B. II only  
 C. Both I and II  
 D. neither I or II

In Figure I, the bond between the two white spheres is broken. In Figure II, six molecules combine to form a solid precipitate. Bond breaking requires energy (endothermic), while bond forming releases energy (exothermic).

4. How many grams of  $\text{N}_2\text{H}_4$  would be required to react completely with 19.0 g of  $\text{O}_2$  according to the balanced chemical equation?



- A. 37.1 g  
 B. 42.1 g  
 C. 74.2 g  
 D. 23.2 g  
 E. 26.5 g

The chemical equation is used to relate the number of moles of one substance to the number of moles of another. If grams of  $\text{O}_2$  is given, it needs to be converted to moles. The coefficients are used as mole ratios, and the formula mass of  $\text{N}_2\text{H}_4$  is used to get back to grams:

$$\begin{aligned} 19.0 \text{ g O}_2 & \left( \frac{1 \text{ mol O}_2}{32.0 \text{ g O}_2} \right) \left( \frac{2 \text{ mol N}_2\text{H}_4}{3 \text{ mol O}_2} \right) \left( \frac{32.0 \text{ g N}_2\text{H}_4}{1 \text{ mol N}_2\text{H}_4} \right) \\ & = 37.3 \text{ g N}_2\text{H}_4 \end{aligned}$$

# General Chemistry Lab Answers

**Linda C. Hodges**



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web checklist for cat excavator inspection oil level equipment daily checklist and safety inspection form engine oil level appearance additional checklist specific to the equipment may be used and attached walk around inspection warning decals check oil level and for leaks forklift inspection checklist

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web jul 2 2018 look for loose bolts trash buildup oil leaks coolant leaks broken parts or worn parts remove accumulated trash dirt grease or oil from the machine for more information please refer to the operation maintenance manual o mm or any other applicable manuals and instructions for this product

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web pre checks are essential to the longevity of your machine check your excavator s fluids regularly and replace them according to the recommended change intervals use your operation and maintenance manual to find and service all grease points don t forget to keep an eye on your excavator s track tension

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web apr 8 2022 every filter on the equipment must have its own check off that says if it was cleaned or replaced that includes oil fuel air and hydraulic filters it could also be cabin filters checklist remarks can include whether filters were cleaned or replaced f analysis of the engine oil and hydraulic oil is mandatory during every maintenance

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web this video walks you through the most important points of excavator maintenance and here s a quick look at six key areas where it pays to focus your efforts 1 undercarriage undercarriage problems can account for more than half of your maintenance costs

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web check oil levels check your equipment s hydraulic and engine oil levels before starting the day check coolant levels

ensure your excavator's coolant levels are adequate and look for possible leaks from your oil cooler and radiator

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daily inspection checklist taking note of the coolant level engine oil select a competent and knowledgeable employee to

regularly inspect heavy equipment belts hoses oil and fluid heavy equipment safety inspection