

Chemistry Benchmark II Study Guide

- Write the formula for the following:
 - Iron (II) oxide
 - calcium phosphate
 - potassium bromide
 - ammonium phosphate
 - copper (II) bromide
 - magnesium hydroxide
 - nitrogen dioxide
 - titanium (IV) oxide
 - xenon hexafluoride
- What is a binary compound? What is a ternary compound?
- Write formulas for
 - hydroiodic acid
 - carbonic acid
 - phosphorous acid
 - sulfuric acid
 - hydrosulfuric acid
 - hydrochloric acid
 - chloric acid
 - chlorous acid
 - hypochlorous acid
 - perchloric acid
- Name these:
 - NaCl
 - KI
 - CsBr
 - Rb₂O
 - SrI₂
 - CuCl
 - HgO
 - Fe₂O₃
 - MnO₂
 - PbCl₄
 - CuI
 - SnBr₄
 - BF₃
 - NO
 - N₂O₅
 - CCl₄
 - P₂O₅
 - N₂O₃
- Find the oxidation number assigned to:
 - S in H₂SO₄
 - P in PCl₅
 - S in SO₃
- Name these.
 - NO₂
 - N₂O
 - SO₃
- How many valence electrons do elements have in the following groups?
 - alkali metals
 - alkaline earth metals
 - halogens
 - noble gases
- Write the complete electron configuration, noble gas configuration, Lewis dot structure, and orbital notation for:
 - Sr
 - Br
 - Na
 - K
 - N
 - Cl
- In each group, identify the atom with the largest atomic radius.
 - Sn, Xe, Rb, Sr
 - Rn, He, Xe, Kr
 - Pb, Ba, Cs, At
- In each group, identify the atom with the highest ionization energy.
 - Cs, K, Li
 - Ba, Sr, Ca
 - I, Br, Cl
 - Mg, Si, S
- Define the following: ionic bond, covalent bond, polar covalent bond, nonpolar covalent bond
- Define electronegativity and describe the trend across a period and down a group. What type of element has the highest electronegativity? What type has the lowest electronegativity?
- Identify the type of bond in each:
 - K-N
 - Cs-O
 - C-N
 - O-F
 - Na-Cl
 - Cl-F
 - Cl-O
 - Ca-Cl
- Who first arranged the periodic table in order of increasing atomic mass?
- Draw Lewis structures for the following and predict the molecular shape.
 - H₂S
 - PH₃
 - O₂
 - sulfate ion
 - SO₂
 - BF₃
- Calculate the wavelength and energy of 96.6MHz radio waves.
- What is the frequency and energy of electromagnetic radiation having a wavelength of $3.33 \times 10^{-6}\text{m}$?
- Predict the ions that the following elements might form:
 - F
 - K
 - I
 - Ga
 - Fe (trick question)
 - P
 - Ag

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N Colangelo



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Dhanasekaran, Viney Chawla, 2024-10-07 Computational methods and understanding computational models are important in modern drug discovery The book focuses on computational approaches that can improve the development of in silico methodologies It includes lead hit methods docking algorithms computational chiral compounds structure based drug design GROMACS and NAMD structural genomics toxicity prediction enzyme inhibitors and peptidomimetic therapeutics

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Australian national bibliography ,1961 **Nanoelectrochemistry** Michael V. Mirkin, Shigeru Amemiya, 2015-03-27 Nanoscale electrochemistry has revolutionized electrochemical research and technologies and has impacted other fields including nanotechnology and nanoscience biology and materials chemistry This book examines well established concepts and principles and provides an updated overview of the field and its applications The first two chapters contain theoretical background specifically theories of electron transfer transport and double layer processes at nanoscale electrochemical interfaces The next chapters examine the electrochemical studies of nanomaterials and nanosystems as well as the applications of nanoelectrochemical techniques Each chapter can be read independently providing readers with a compact up to date review of the ERDA Energy Research Abstracts United States. Energy Research and Development

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