

## General Chemistry placement test Study Guide 2024 Solved 100%

**Atomic Number** - The number of protons in the nucleus of an atom. The atomic number of an element never changes, meaning that the number of protons in the nucleus of every atom in an element is always the same.

Hydrogen = 1, Carbon = 6, Oxygen = 8 (located on top of element of periodic table)

**Mass Number** - # of neutrons + # of protons = mass number (located on bottom of periodic table)

**Isotopes** - Let's say an atom is missing a neutron or has an extra neutron. That type of atom is called an isotope. An atom is still the same element if it is missing an electron.

Example: (carbon-14)

**Distribution &/W subatomic particles in an Atom** - A typical atom consists of three subatomic particles: Protons, Neutrons, and Electrons.

- Nucleons include protons and neutrons. All the positive charge of an atom is contained in the nucleus, and originates from the protons. Neutrons are neutrally-charged. Electrons, which are negatively-charged, are located outside of the nucleus.

**Roles of subatomic particles: Protons, Neutrons, & Electrons** - **Particle Mass:**

Protons & Neutrons have about same mass. Electrons have much less mass than protons and neutrons.

**Particle Charge:**

Protons have a positive charge (+)

- Neutrons have no charge (0)

- Net charge of the nucleus is positive and equal to the number of protons.

# General Chemistry Assessment Test Study Guide

**David Kirk**



## **General Chemistry Assessment Test Study Guide:**

## **General Chemistry Assessment Test Study Guide** Book Review: Unveiling the Power of Words

In a global driven by information and connectivity, the power of words has be much more evident than ever. They have the capability to inspire, provoke, and ignite change. Such may be the essence of the book **General Chemistry Assessment Test Study Guide**, a literary masterpiece that delves deep into the significance of words and their affect our lives. Written by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book is key themes, examine its writing style, and analyze its overall affect readers.

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