

 **$K_{sp}$  and ICE Problems Explained**

**Key Point:**  
**Unsolvable System**

- a. Write initial concentrations and write solubility product expression according to the equation:



Initial	$K_{sp}$	Change	Equilibrium
$\text{PbCl}_2$	$1.7 \times 10^{-2}$	$-x$	$1.7 \times 10^{-2} - x$
$\text{Pb}^{2+}$		$+x$	$x$
$\text{Cl}^-$		$+2x$	$2x$

- b. Write initial concentrations and write solubility product expression according to the equation:



Initial	$K_{sp}$	Change	Equilibrium
$\text{PbCl}_2$	$1.7 \times 10^{-2}$	$-x$	$1.7 \times 10^{-2} - x$
$\text{Pb}^{2+}$		$+x$	$x$
$\text{Cl}^-$		$+2x$	$2x$

- c. Use initial  $K_{sp}$  to obtain the equation to solve and solve for concentration and solubility. Write your answer (b) after it is solved for the concentration:

$$K_{sp} = 1.7 \times 10^{-2} = [\text{Pb}^{2+}][\text{Cl}^-]^2 = (x)(2x)^2 = 4x^3$$

**If the system is unsolvable, write the equation to solve for the concentration and solubility. Write your answer (b) after it is solved for the concentration:**

Initial	$K_{sp}$	Change	Equilibrium	Initial	Equilibrium
$\text{PbCl}_2$	$1.7 \times 10^{-2}$	$-x$	$1.7 \times 10^{-2} - x$	$\text{Pb}^{2+}$	$x$
$\text{Pb}^{2+}$		$+x$	$x$	$\text{Cl}^-$	$2x$
$\text{Cl}^-$		$+2x$	$2x$	$\text{PbCl}_2$	$1.7 \times 10^{-2} - x$

# Equilibrium Ice Box Answers

**JS Bruner**



## **Equilibrium Ice Box Answers:**

## Decoding **Equilibrium Ice Box Answers**: Revealing the Captivating Potential of Verbal Expression

In a period characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its power to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Equilibrium Ice Box Answers**," a mesmerizing literary creation penned by way of a celebrated wordsmith, readers set about an enlightening odyssey, unraveling the intricate significance of language and its enduring impact on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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