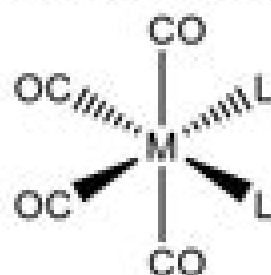
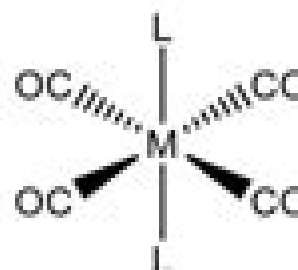


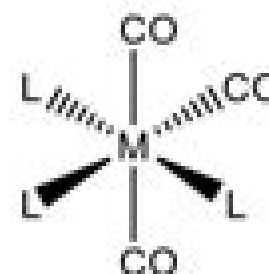
1. Completed online (Molecular symmetry (Group theory) Quiz) [4]
2. Completed online (Molecular symmetry (Group theory) Quiz) [6]
3.  $M(CO)_6$  (where  $M$  = transition metal) has been utilised in a reaction. The product was analysed by infra-red and Raman spectroscopy. The CO ligands of the product produced three frequencies in the infra-red spectrum and three frequencies in the Raman spectrum. The infra-red and Raman frequencies were co-incident. The possible structures (A-C) and their point groups (in parentheses) are shown below. Deduce, using a group theory approach, which of the compounds has been synthesised.



A ( $C_{2v}$ )



B ( $D_{4h}$ )



C ( $C_{2v}$ )

N.B. The identity of the ligand, L, can be taken to be a single atom to simplify analysis.

You must include the following in your answer:

- The reducible representation of  $\Gamma_{M-CO}$  for each complex [9]
- The irreducible representation of  $\Gamma_{M-CO}$  for each complex, stating clearly which labels are infra-red and / or Raman active. The reduction formula must be used (and calculations shown) to deduce the labels present for each complex [12]
- Which complex has been synthesised and brief reasoning for your answer [2]

# Molecular Symmetry Group Theory Answers To

**N Noddings**



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