Name: Date: Lab Section	Name:	Date:	Lab Section
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# Experiment 2: Molecular Models REPORT SHEET

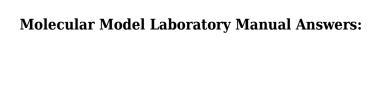
Pre-lab Study Questions:

- Objectives of the experiment (in your own words):
- 2. What kinds of electrons form the covalent bonds in molecules?
- 3. Is a polar molecule symmetrical or asymmetrical?

FORMULA OF MOLECULE	LEWIS STRUCTURE	3-D SHAPE w/lone pairs	MOLECULAR GEOMETRY	MOLECULAR POLARITY
example: CH <sub>4</sub>	H 11—C—H		tetrahedral	nonpolar
H <sub>2</sub> O	H H	H . 111 P:	tet Yako Kal	
H <sup>0</sup> O,	H:0:H	# S H	tethadal	nonpolar
C <sub>2</sub> H <sub>6</sub>				12-7
C <sub>0</sub> H <sub>4</sub>	H C = C H		tiyon	Nen poker
C <sub>2</sub> H <sub>2</sub>	44 6 2			
CO <sub>1</sub>	6 = 6 = 6		Linear	Nanpolar
18 <sub>2</sub> O <sub>2</sub>				
CHICI				

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Katrin Zwirglmaier



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