

## Dilutions Worksheet - Solutions

- 1) If I add 25 mL of water to 125 mL of a 0.15 M NaOH solution, what will the molarity of the diluted solution be?

$$\begin{aligned}M_1V_1 &= M_2V_2 \\(0.15\text{ M})(125\text{ mL}) &= x(150\text{ mL}) \\x &= 0.125\text{ M}\end{aligned}$$

- 2) If I add water to 100 mL of a 0.15 M NaOH solution until the final volume is 150 mL, what will the molarity of the diluted solution be?

$$\begin{aligned}M_1V_1 &= M_2V_2 \\(0.15\text{ M})(100\text{ mL}) &= x(150\text{ mL}) \\x &= 0.100\text{ M}\end{aligned}$$

- 3) How much 0.05 M HCl solution can be made by diluting 250 mL of 10 M HCl?

$$\begin{aligned}M_1V_1 &= M_2V_2 \\(10\text{ M})(250\text{ mL}) &= (0.05\text{ M})x \\x &= 50,000\text{ mL}\end{aligned}$$

- 4) I have 345 mL of a 1.5 M NaCl solution. If I boil the water until the volume of the solution is 250 mL, what will the molarity of the solution be?

$$\begin{aligned}M_1V_1 &= M_2V_2 \\(1.5\text{ M})(345\text{ mL}) &= x(250\text{ mL}) \\x &= 2.07\text{ M}\end{aligned}$$

- 5) How much water would I need to add to 500 mL of a 2.4 M KCl solution to make a 1.0 M solution?

$$\begin{aligned}M_1V_1 &= M_2V_2 \\(2.4\text{ M})(500\text{ mL}) &= (1.0\text{ M})x \\x &= 1200\text{ mL}\end{aligned}$$

1200 mL will be the final volume of the solution. However, since there's already 500 mL of solution present, you only need to add 700 mL of water to get 1200 mL as your final volume. The answer: 700 mL.

# Molarity By Dilution Worksheet Page 69 Answers

**Sebastian Brünink**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

**Molarity By Dilution Worksheet Page 69 Answers:**

## Reviewing **Molarity By Dilution Worksheet Page 69 Answers**: Unlocking the Spellbinding Force of Linguistics

In a fast-paced world fueled by information and interconnectivity, the spellbinding force of linguistics has acquired newfound prominence. Its capacity to evoke emotions, stimulate contemplation, and stimulate metamorphosis is actually astonishing. Within the pages of "**Molarity By Dilution Worksheet Page 69 Answers**," an enthralling opus penned by a highly acclaimed wordsmith, readers embark on an immersive expedition to unravel the intricate significance of language and its indelible imprint on our lives. Throughout this assessment, we shall delve in to the book is central motifs, appraise its distinctive narrative style, and gauge its overarching influence on the minds of its readers.

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