

Molar Mass Worksheet #1

Calculate the molar masses of the following chemicals:

1) Cl_2 $2(35.45) = \boxed{70.9 \text{ g/mol}}$

2) KOH $39.10 + 16.00 + 1.01 = \boxed{56.11 \text{ g/mol}}$

3) BeCl_2 $9.01 + 2(35.45) = \boxed{79.91 \text{ g/mol}}$

4) FeCl_3 $55.85 + 3(35.45) = \boxed{162.2 \text{ g/mol}}$

5) BF_3 $10.81 + 3(19.00) = \boxed{67.81 \text{ g/mol}}$

6) CCl_2F_2 $12.01 + 2(35.45) + 2(19.00) = \boxed{120.91 \text{ g/mol}}$

7) Mg(OH)_2 $24.31 + 2(16.00) + 2(1.01) = \boxed{58.33 \text{ g/mol}}$

8) H_2P_6 $238.03 + 6(19.00) = \boxed{352.03 \text{ g/mol}}$

9) SO_2 $32.07 + 2(16.00) = \boxed{64.07 \text{ g/mol}}$

10) H_3PO_4 $3(1.01) + 30.97 + 4(16.00) = \boxed{98.00 \text{ g/mol}}$

11) $(\text{NH}_4)_2\text{SO}_4$ $2(14.01) + 8(1.01) + 32.07 + 4(16.00) = \boxed{132.17 \text{ g/mol}}$

12) CH_3COOH $2(12.01) + 4(1.01) + 2(16.00) = \boxed{60.06 \text{ g/mol}}$

13) $\text{Pb(NO}_3)_2$ $207.2 + 2(14.01) + 6(16.00) = \boxed{331.22 \text{ g/mol}}$

14) $\text{Ga}_2(\text{SO}_4)_3$ $2(69.72) + 3(32.07) + 9(16.00) = \boxed{379.65 \text{ g/mol}}$

Molar Mass Formula Answer Key

Didier Musso



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